

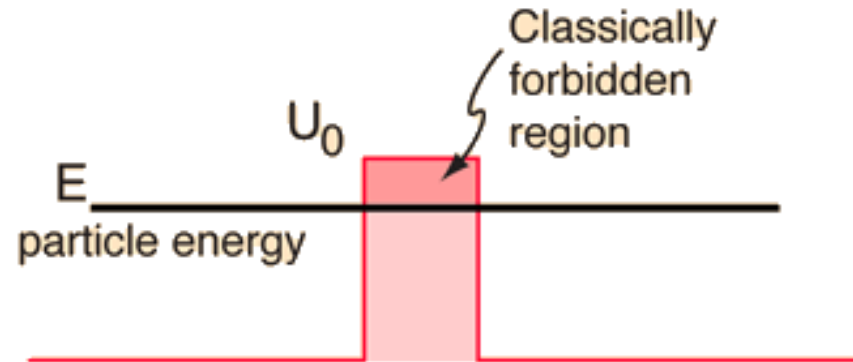
Tunneling Current Increasing with Barrier Width explained using a 2-barrier Model

Saurabh Soni

Zernike Institute of Advanced Materials &
Stratingh Institute for Chemistry
University of Groningen, The Netherlands

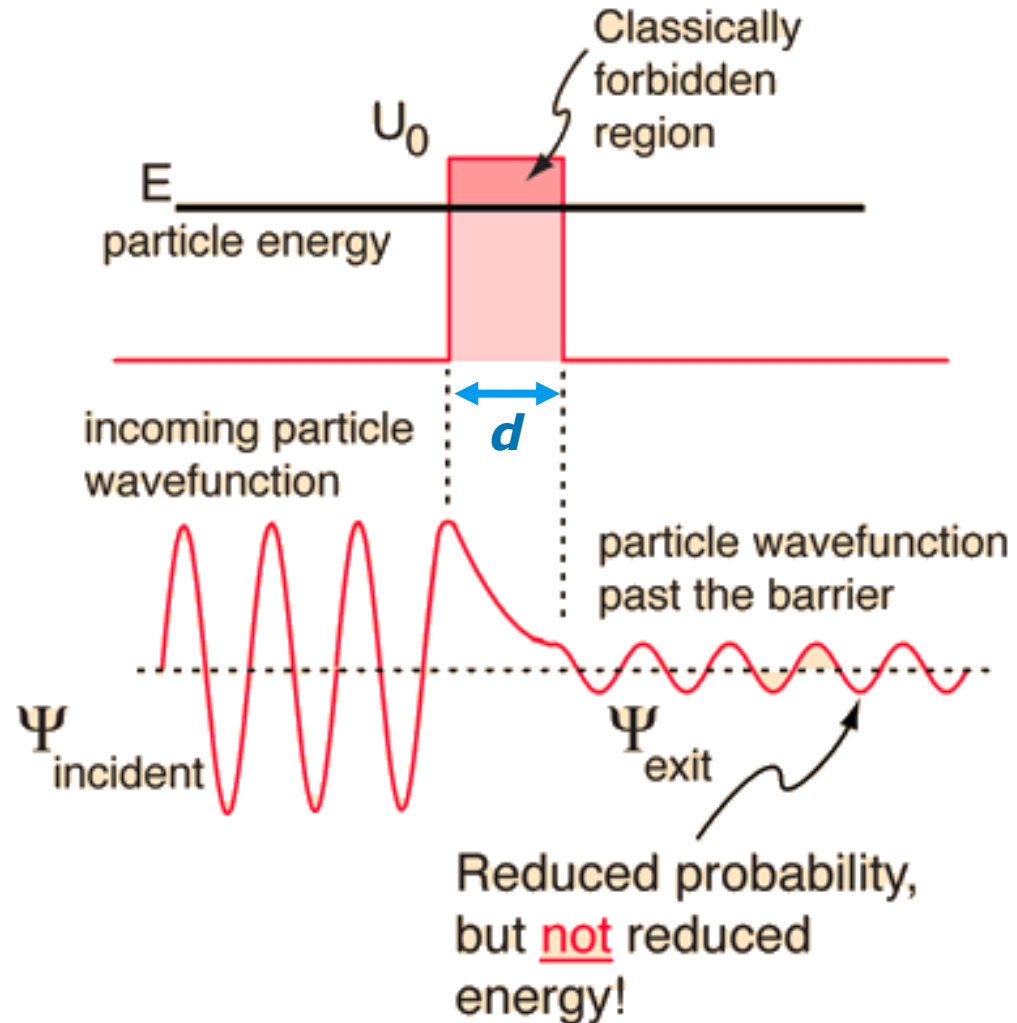
We all learnt in kindergarten...

Tunnelling through a rectangular barrier...



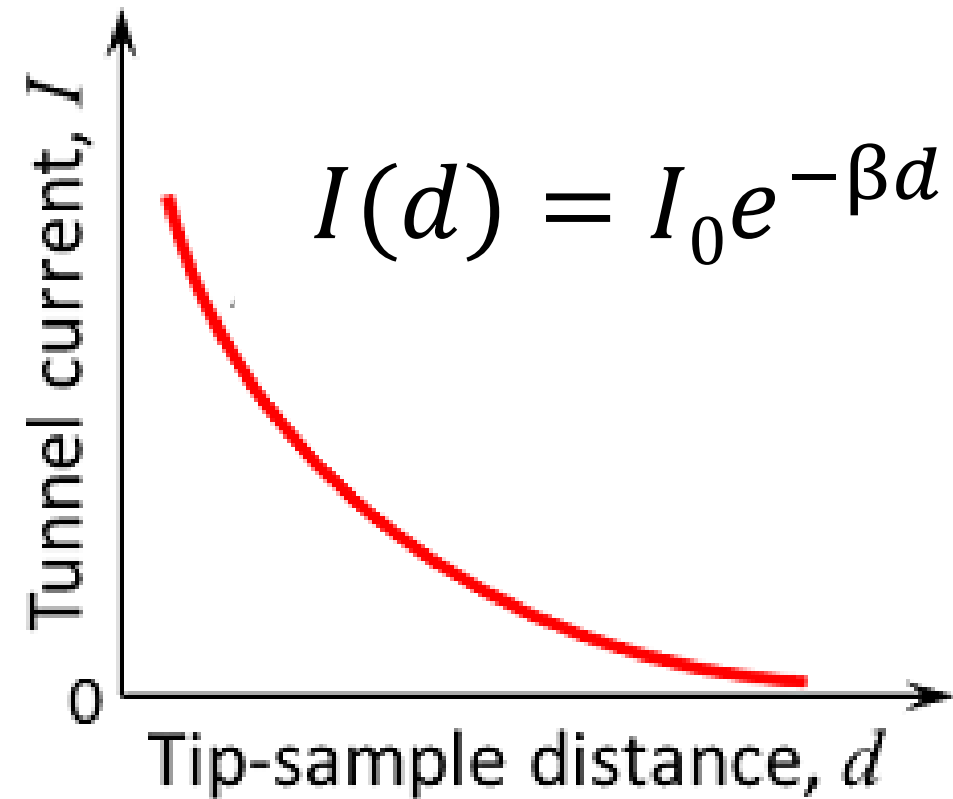
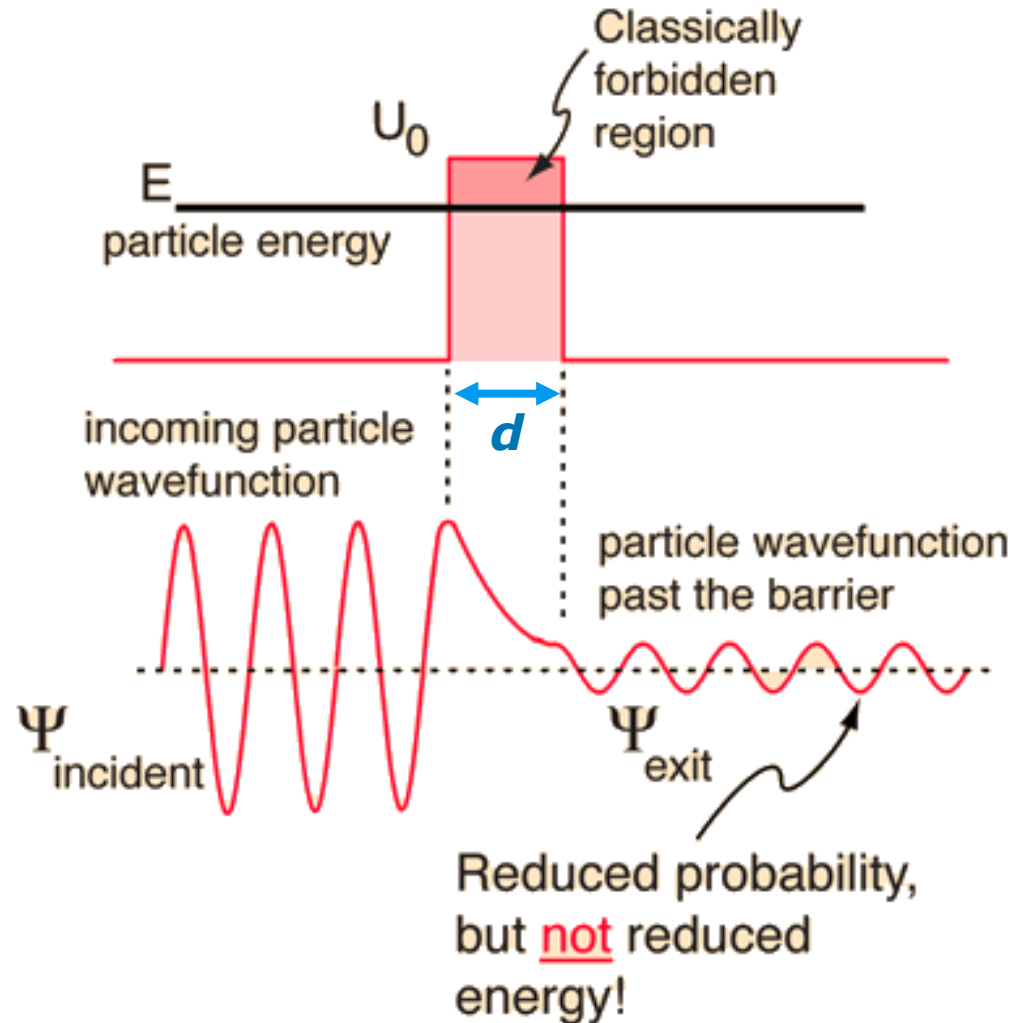
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Tunnelling through a rectangular barrier...



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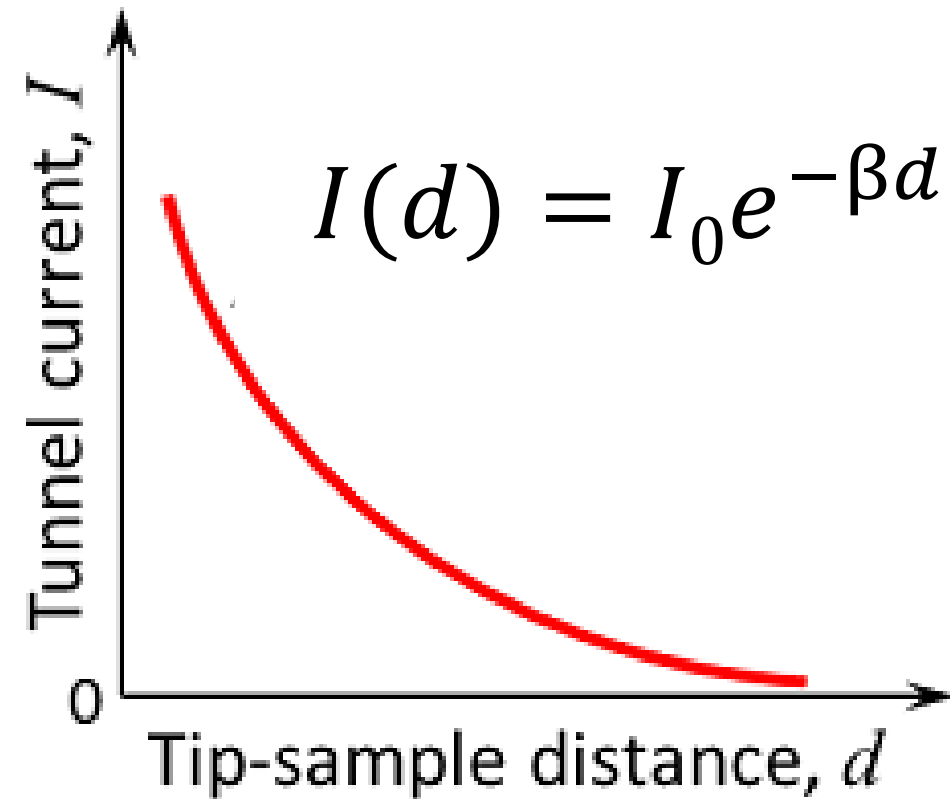
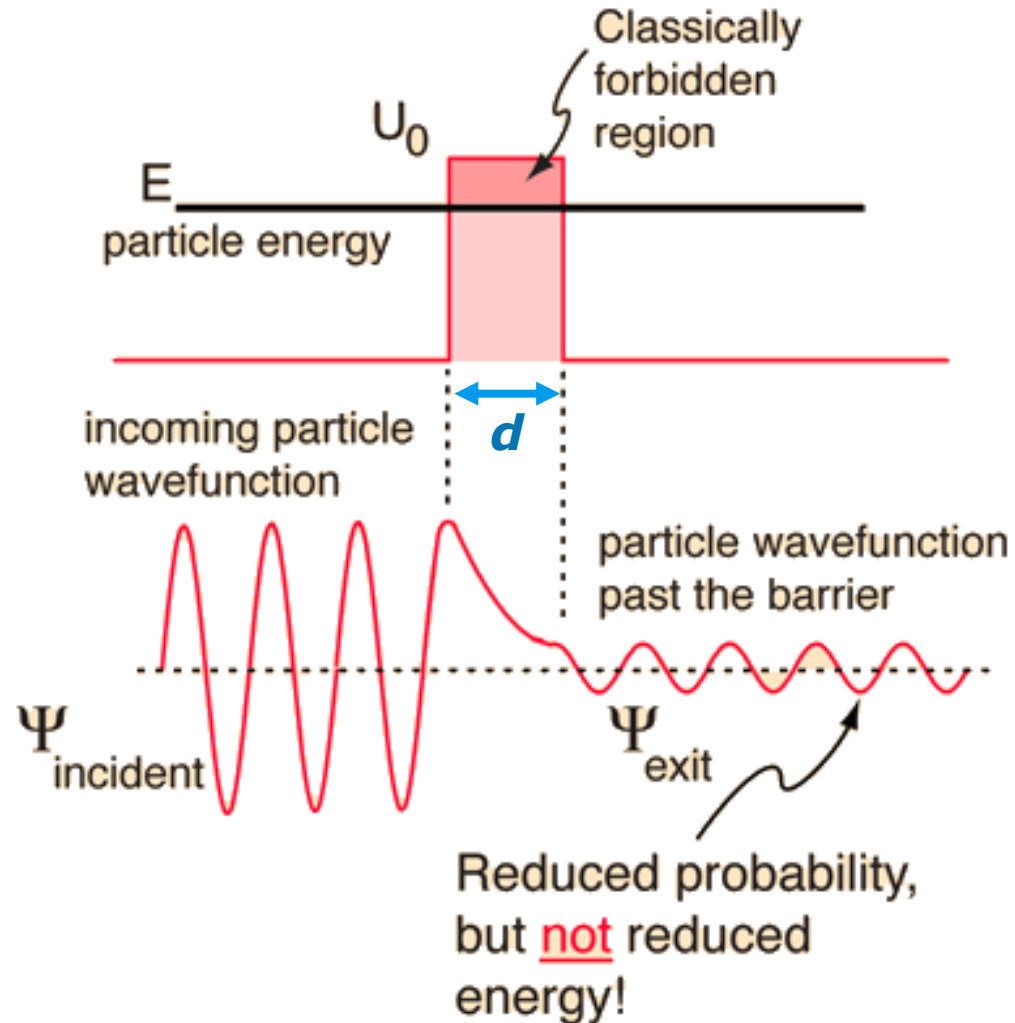
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We all learnt in kindergarten...

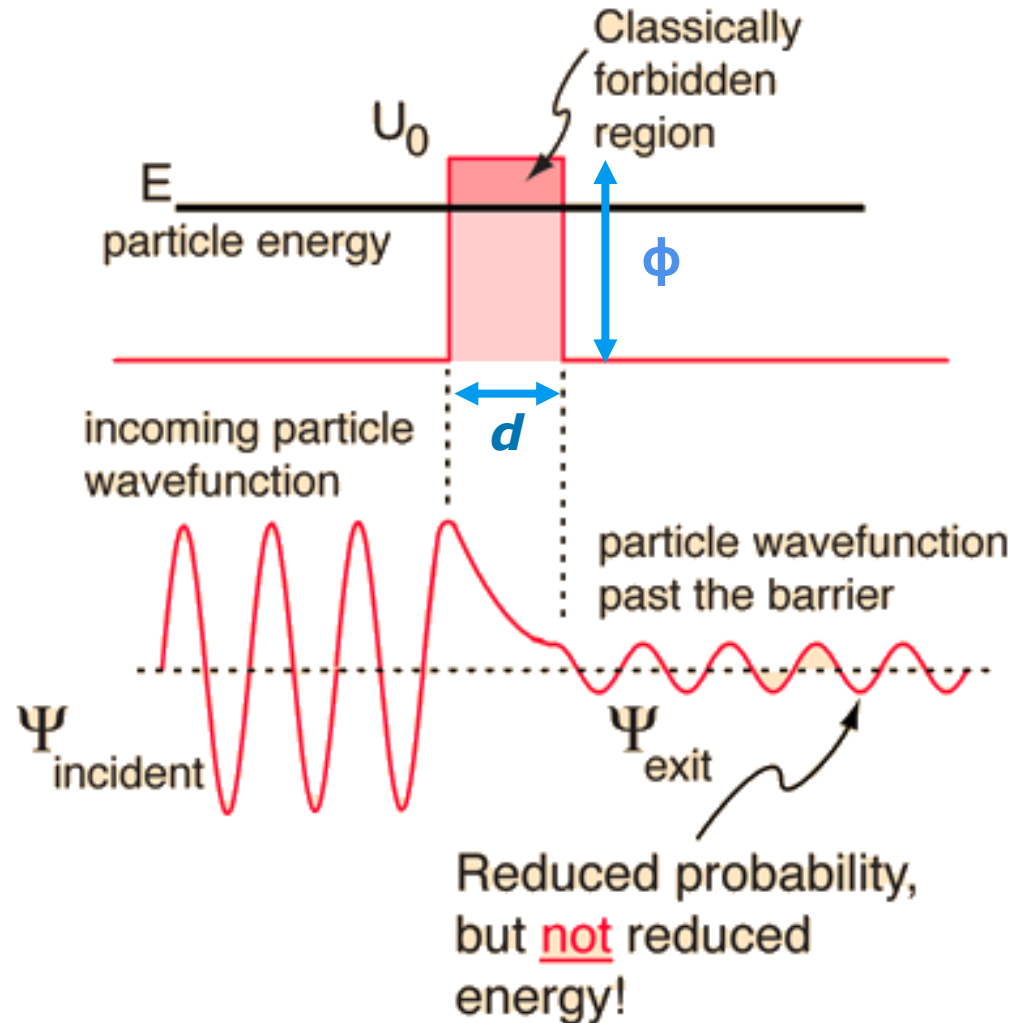
Tunnelling through a rectangular barrier...

$\beta = \text{constant}$



We all learnt in kindergarten...

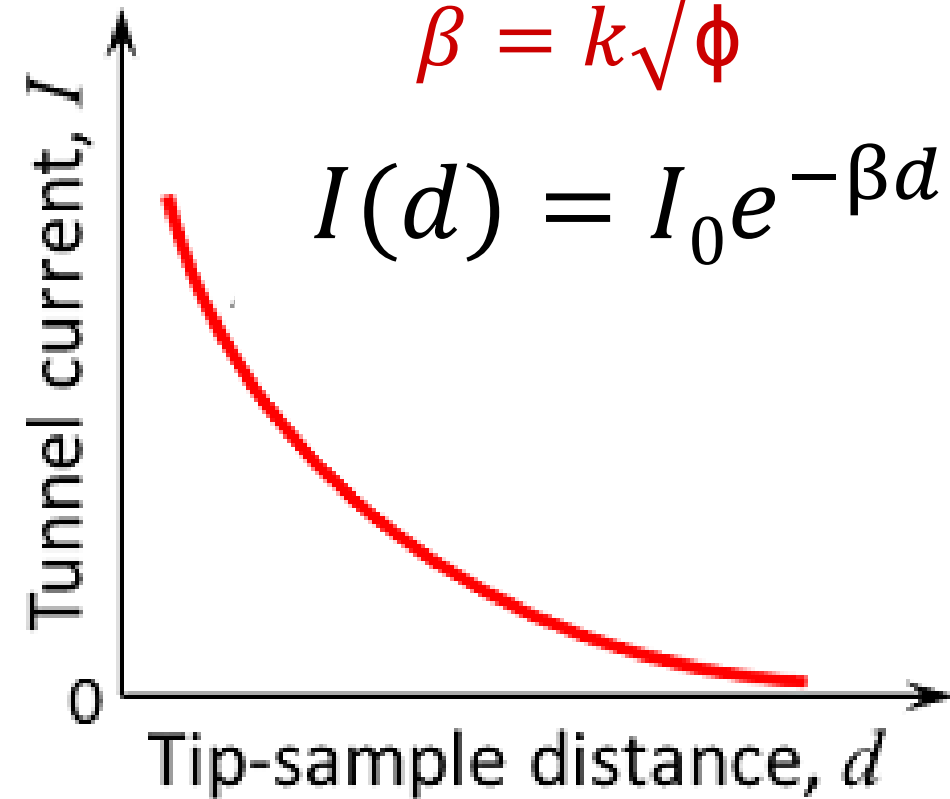
Tunnelling through a rectangular barrier...



$\beta = \text{constant}$

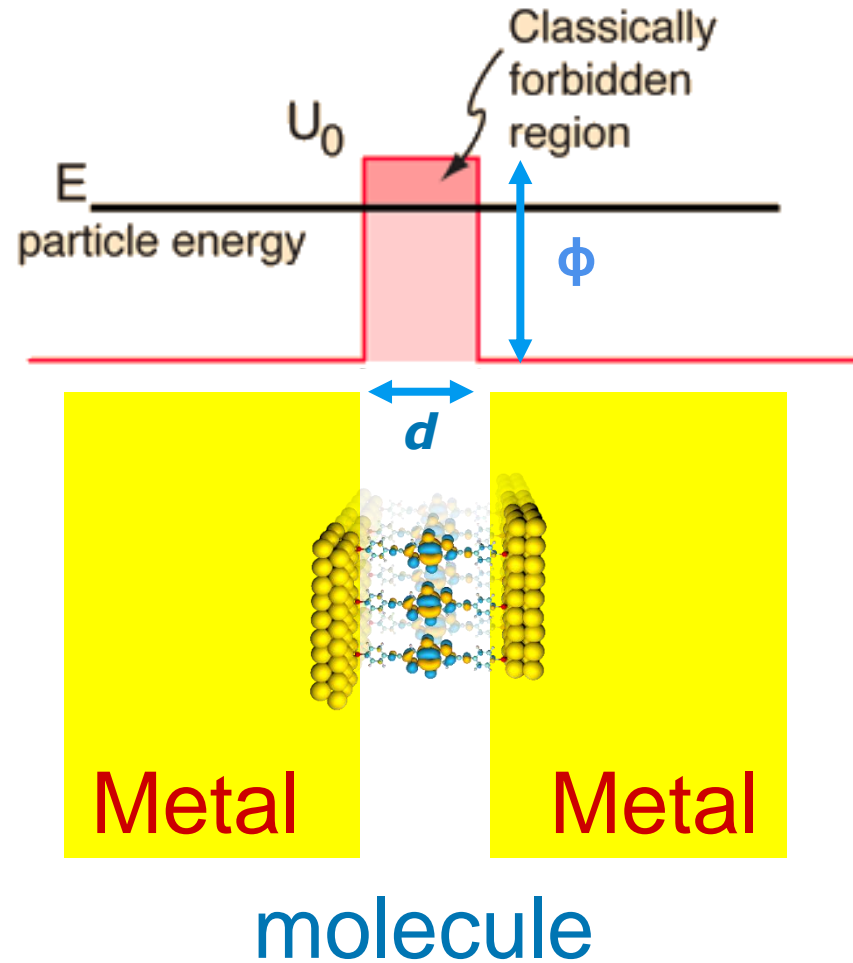
$$\beta = k\sqrt{\phi}$$

$$I(d) = I_0 e^{-\beta d}$$



Applying in Molecular Electronics...

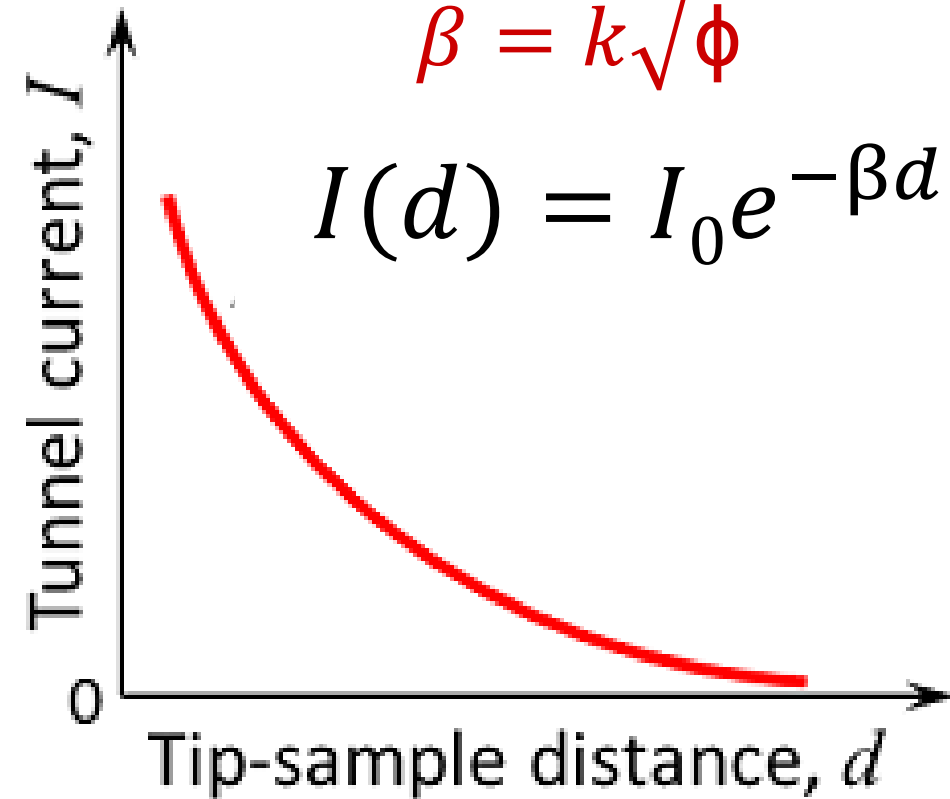
Tunnelling through a molecular barrier...



$\beta = \text{constant}$

$$\beta = k\sqrt{\phi}$$

$$I(d) = I_0 e^{-\beta d}$$



Literature examples

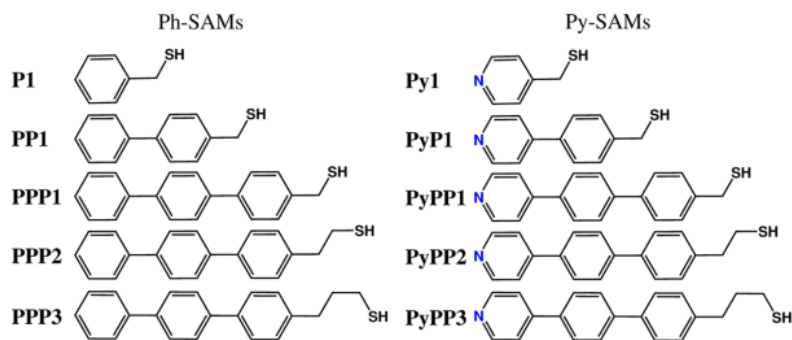
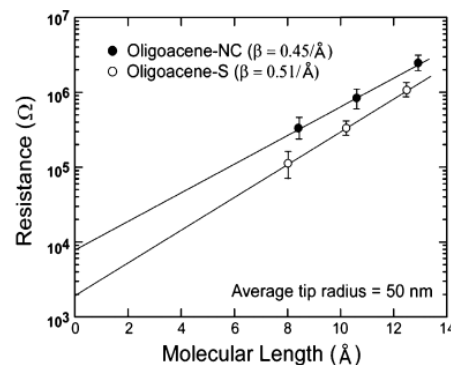
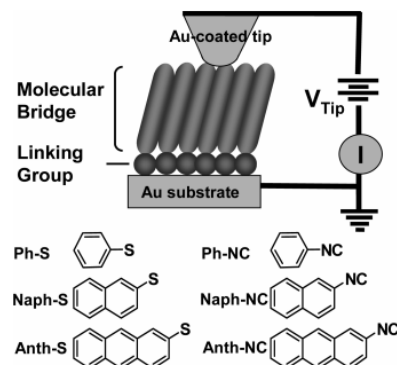
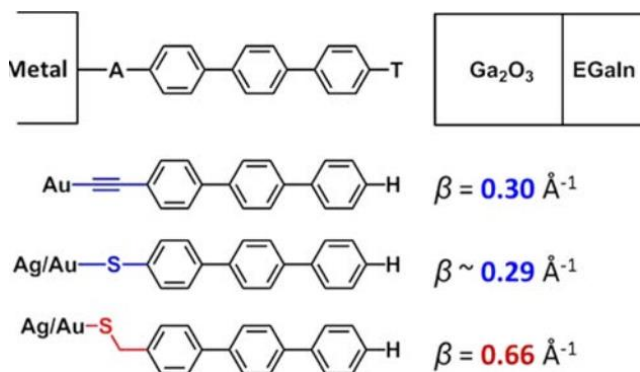


Table 1. Values of β and J_0 for Subsets and Combinations of the Ph-SAM and Py-SAM Series

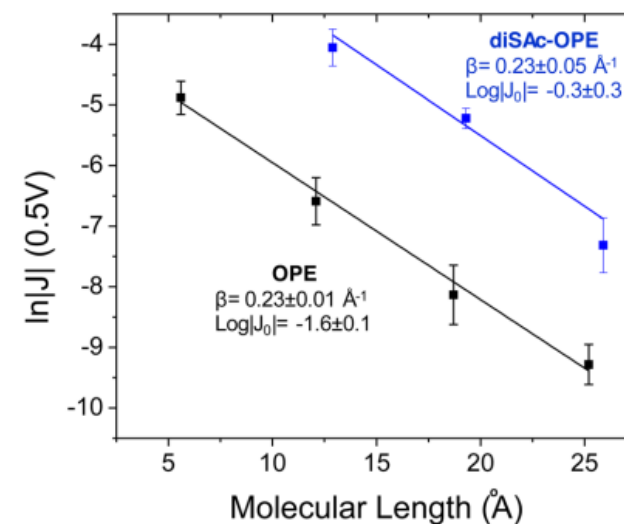
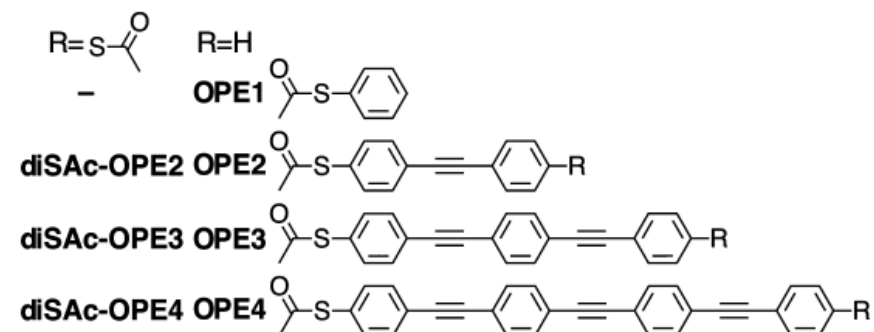
trend	molecules	β (\AA^{-1})	J_0 (A/cm^2)
full series	Ph-SAM	0.44 ± 0.04	1.18 ± 1.44
	Py-SAM	0.42 ± 0.08	2.46 ± 3.00
odd only	P1, PP1, PPP1, PPP3	0.47 ± 0.04	1.45 ± 1.43
	Py1, PyP1, PyPP1, PyPP3	0.40 ± 0.07	2.35 ± 2.77
aryl	P1, PP1, PPP1	0.46 ± 0.07	1.32 ± 1.85
	Py1, PyP1, PyPP1	0.32 ± 0.17	1.11 ± 6.00
alkyl	PPP1, PPP2, PPP3	0.56 ± 0.41	n.d.
	PyPP1, PyPP2, PyPP3	0.36 ± 0.41	n.d.

J. Phys. Chem. C 2013, 117, 11367–11376



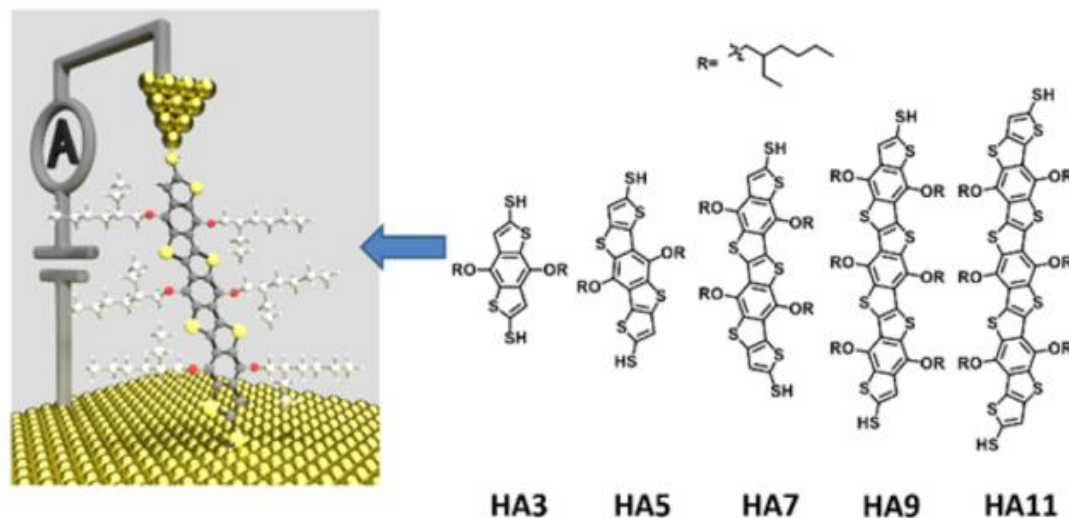
J. AM. CHEM. SOC. 2006, 128, 4970–4971

$\beta = \text{constant}$

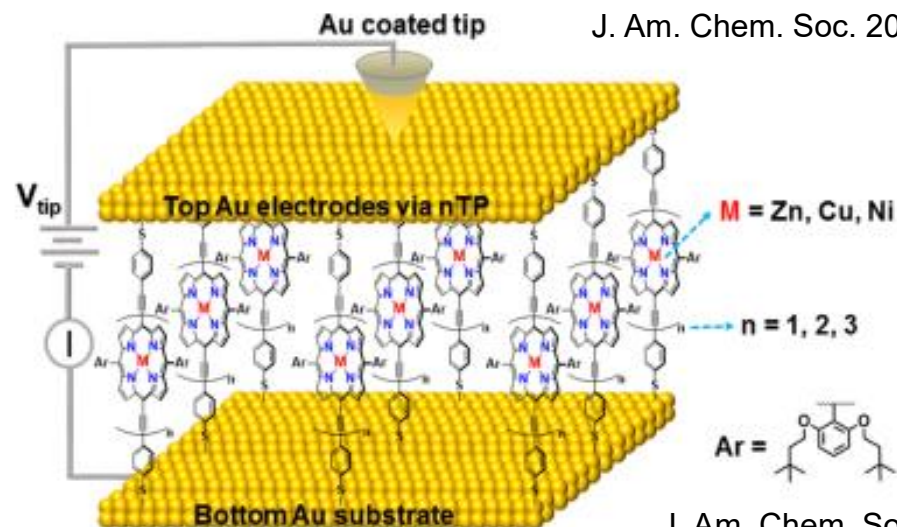


J. Phys. Chem. C 2016, 120, 20437–20445

Literature examples

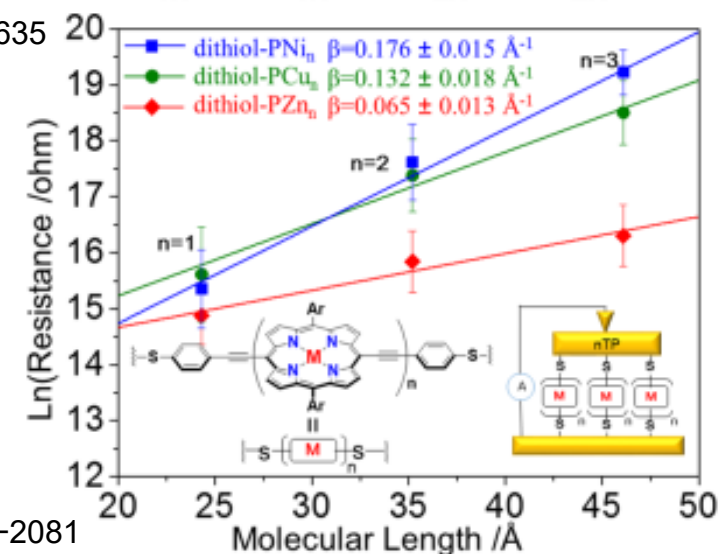
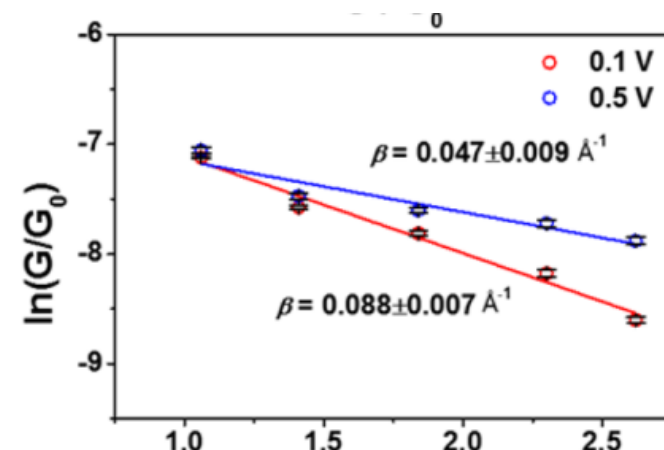


J. Am. Chem. Soc. 2016, 138, 10630–10635

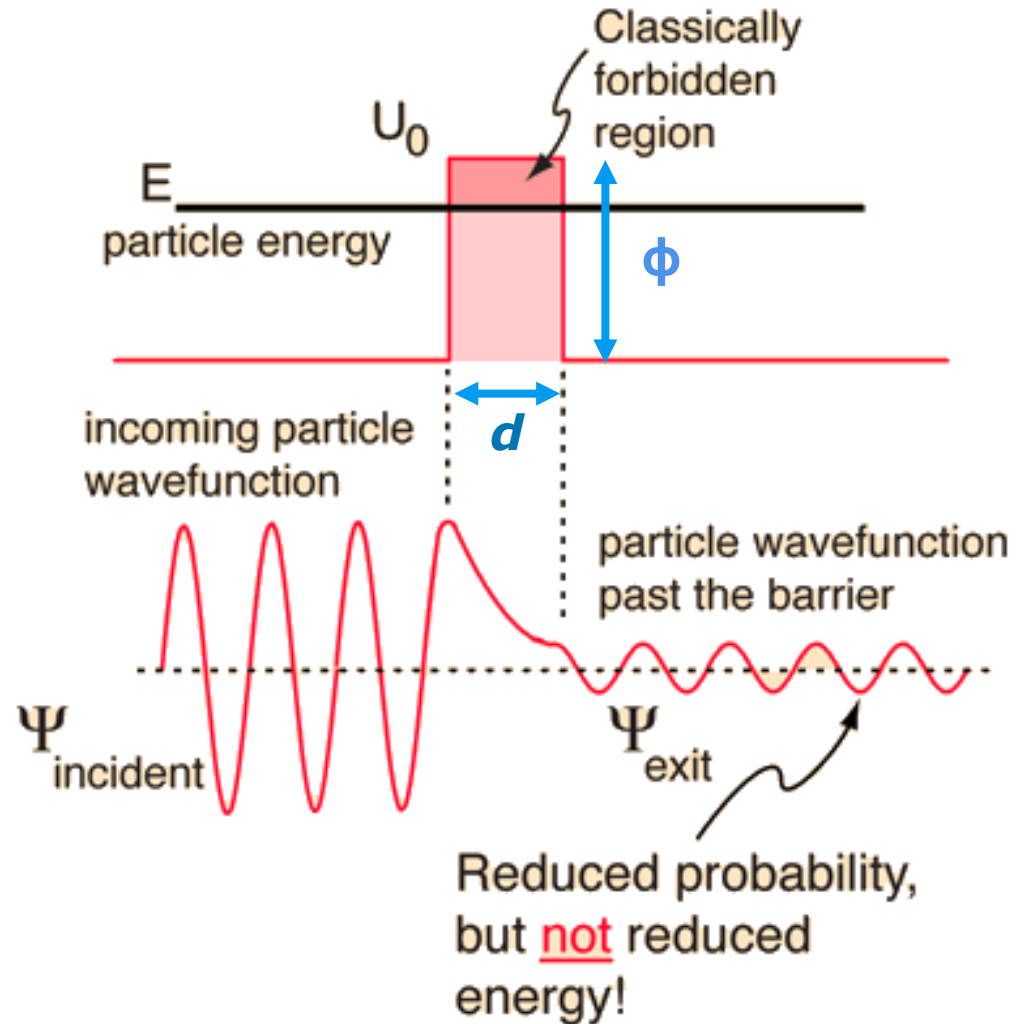


J. Am. Chem. Soc. 2016, 138, 2078–2081

$$\beta \sim 0$$

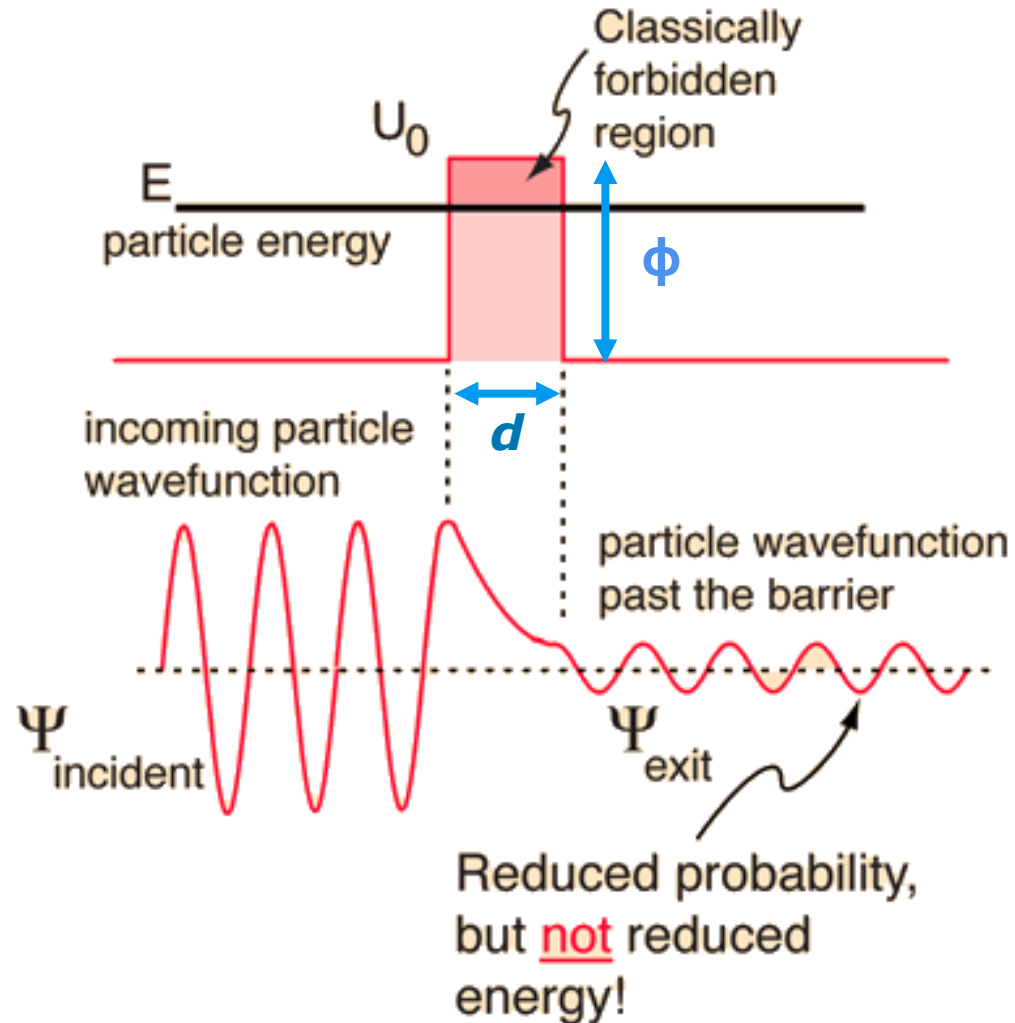


Common Claim!



$$\beta = \text{constant}$$

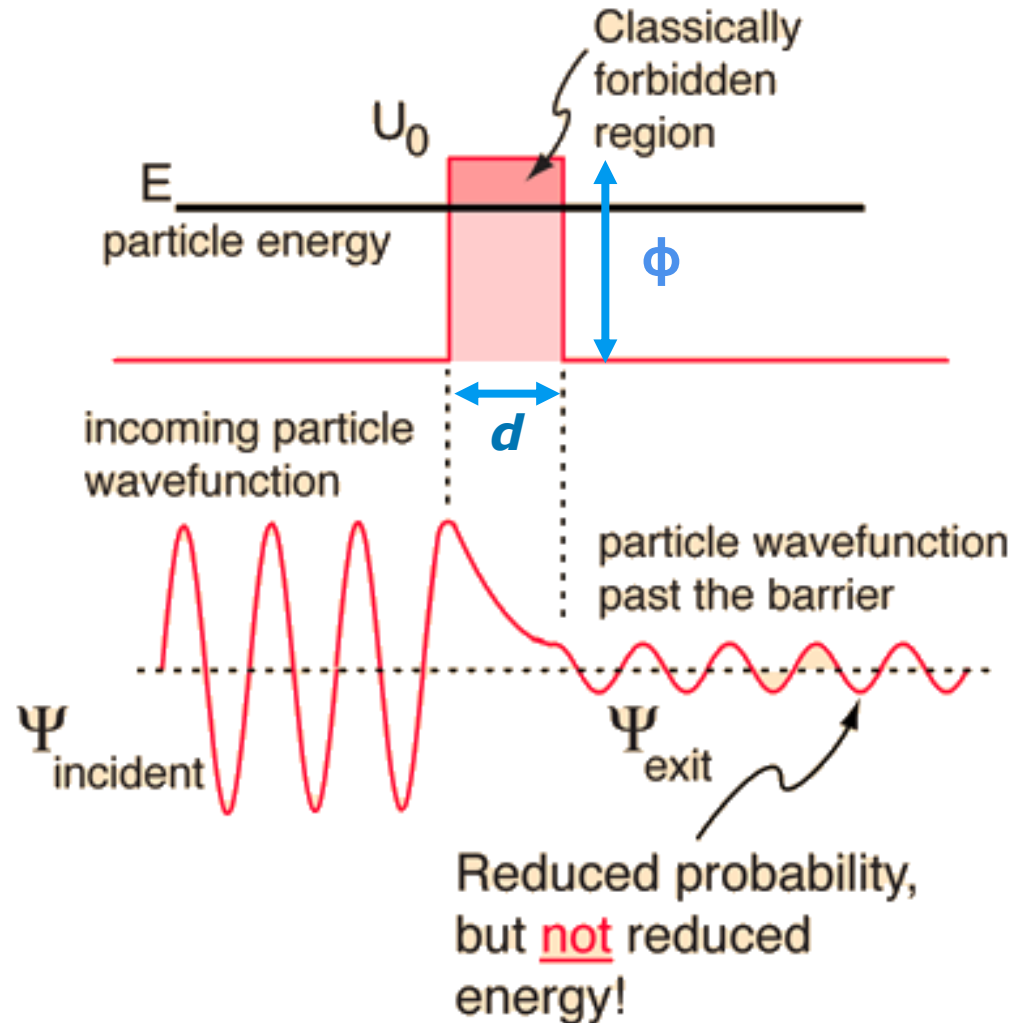
Common Assumption!



$$\beta = \text{constant}$$

$$\rightarrow \phi = \text{constant}$$

Common Assumption!



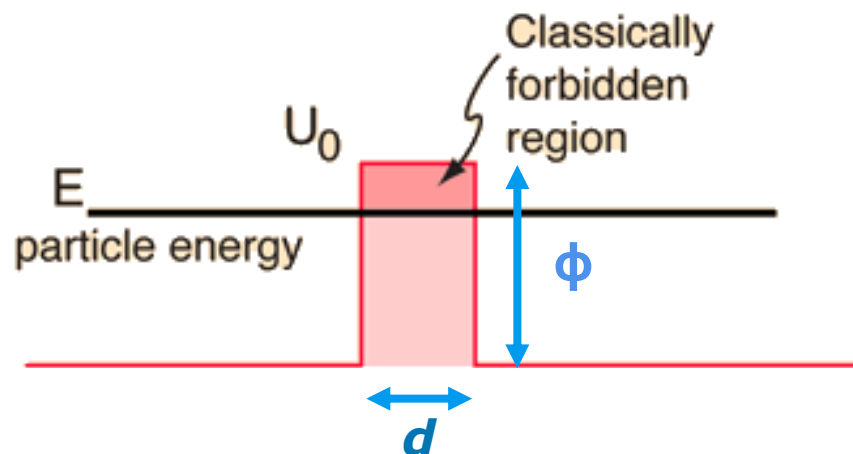
$$\beta = \text{constant}$$

$$\rightarrow \phi = \text{constant}$$

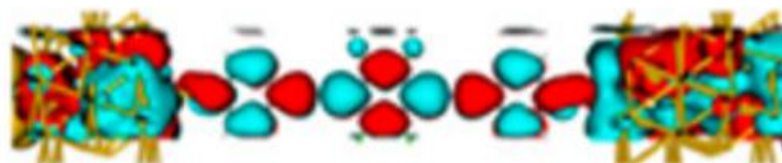
But we know E_g changes!

$$I(d) = I_0 e^{-\beta d}$$

Common Assumption!



- So there should be **fermi-pinning**
 - i.e., $(E_{\text{HOMO/LUMO}} - E_{\text{Fermi}}) = \text{constant}$
 - hybridization between the molecule and the electrodes



$$\beta = \text{constant}$$

$$\rightarrow \phi = \text{constant}$$

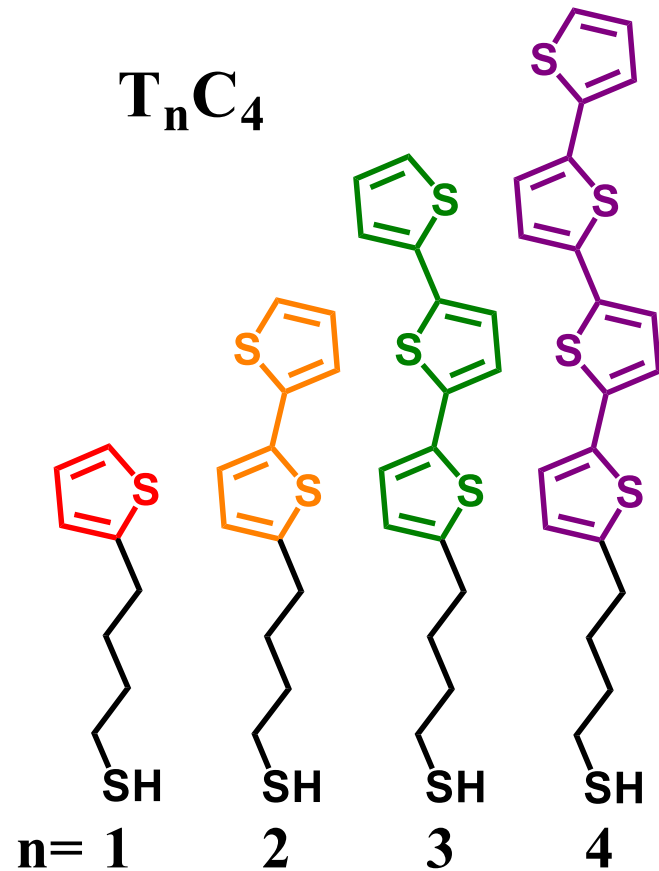
So ϕ would change!

$$I(d) = I_0 e^{-\beta d}$$

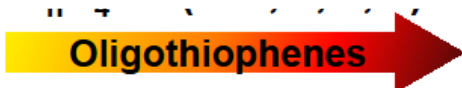
Goals!

- Exclude the effect of **Fermi pinning**
- Understand the influence of barrier height on current

Target system

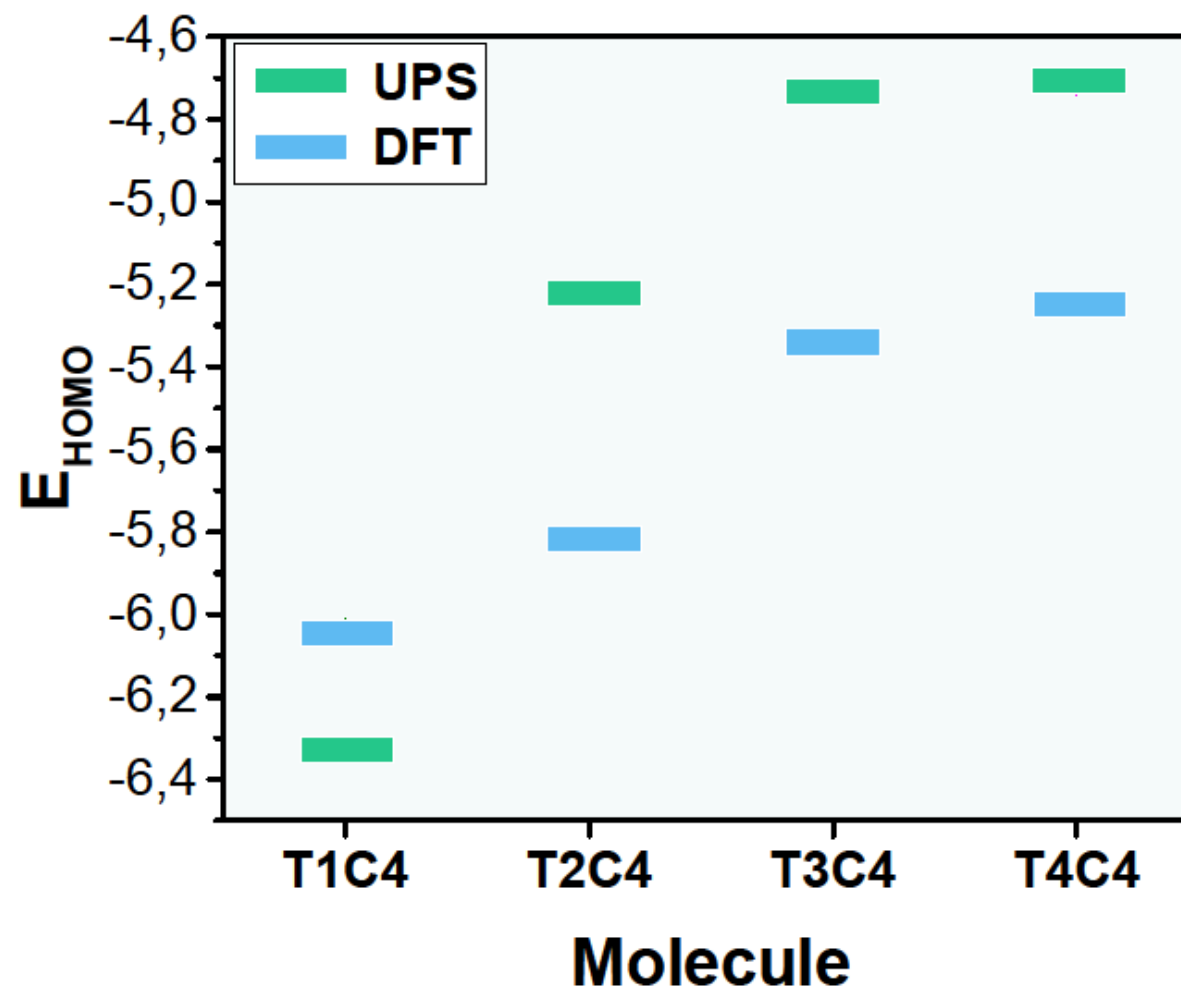
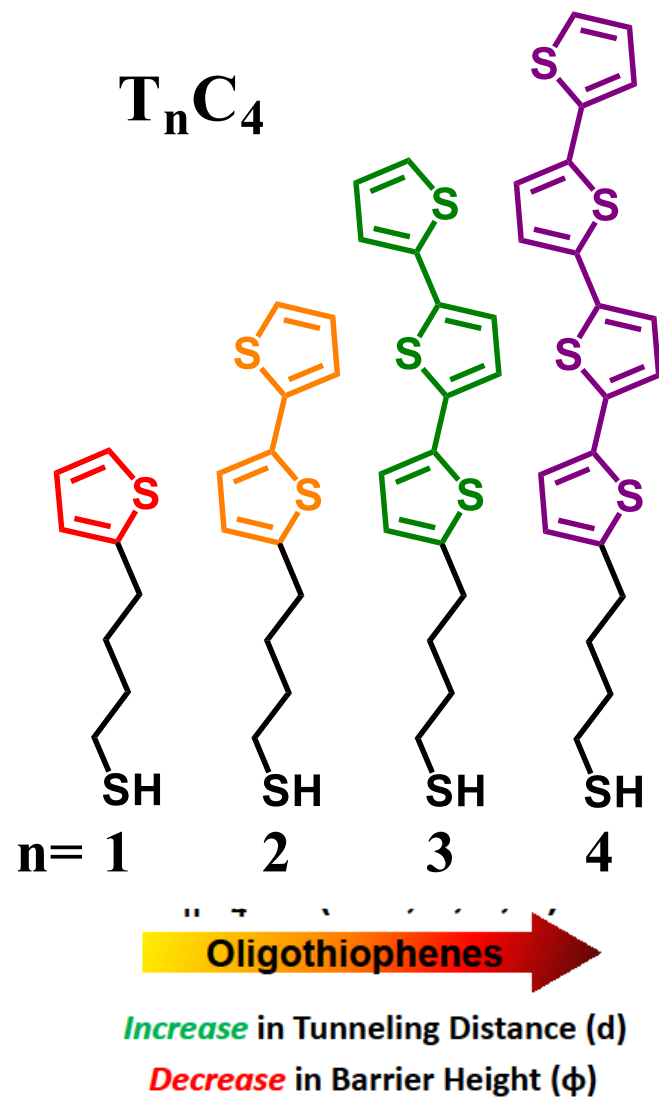


Oligothiophene series


 Oligothiophenes

Increase in Tunneling Distance (d)
Decrease in Barrier Height (ϕ)

Target system: Oligothiophene series



Why Oligothiophene?

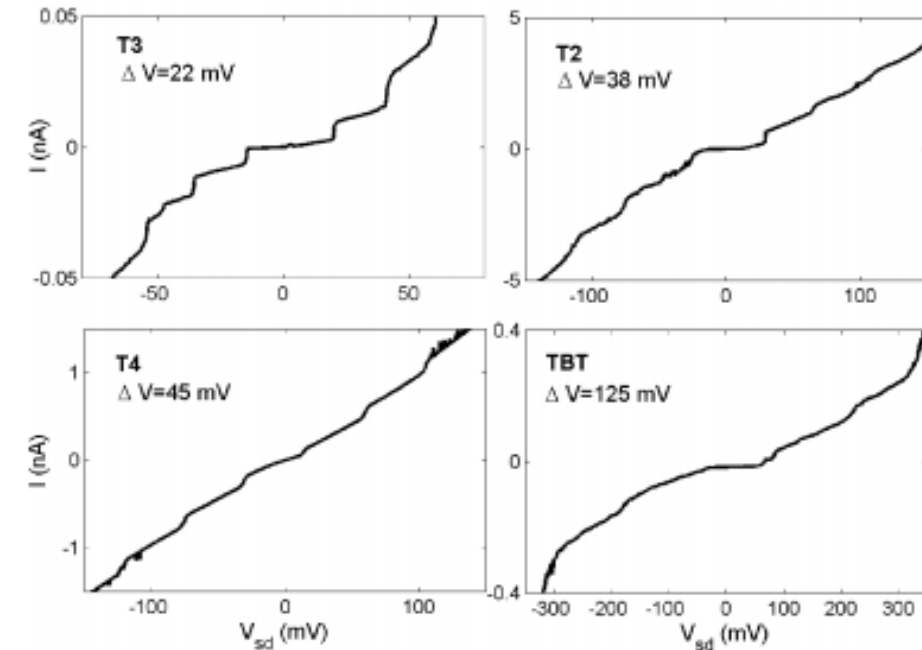
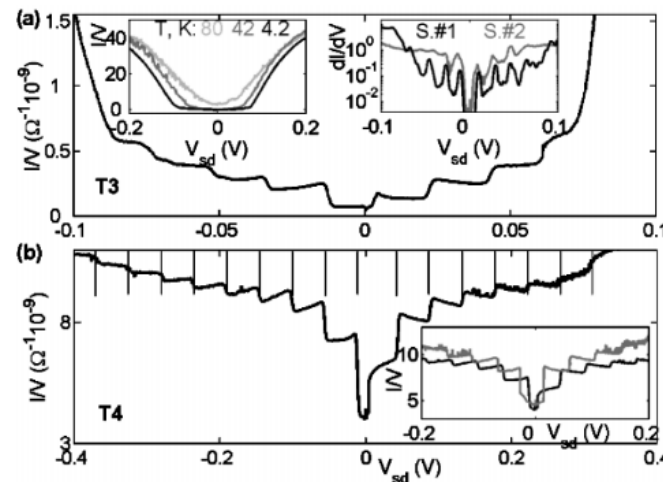
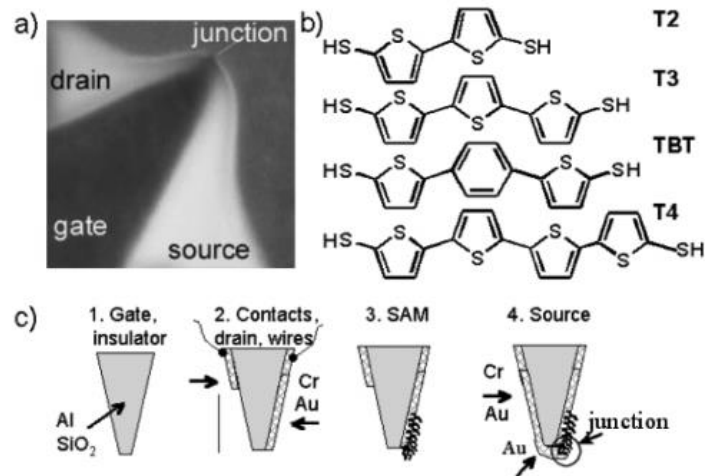
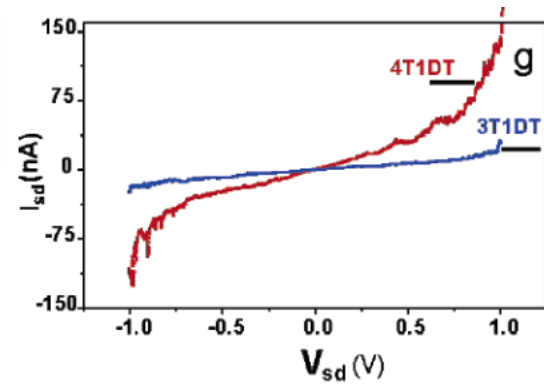
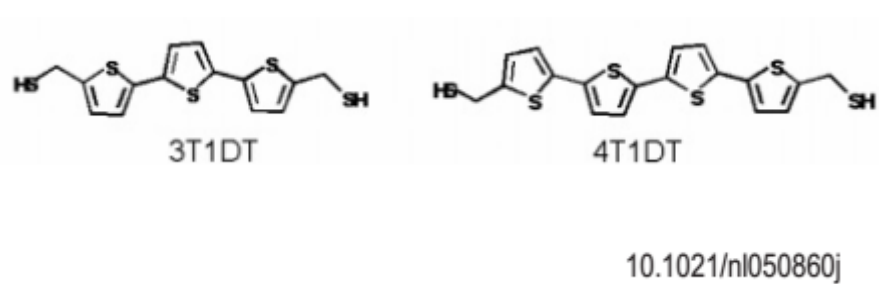
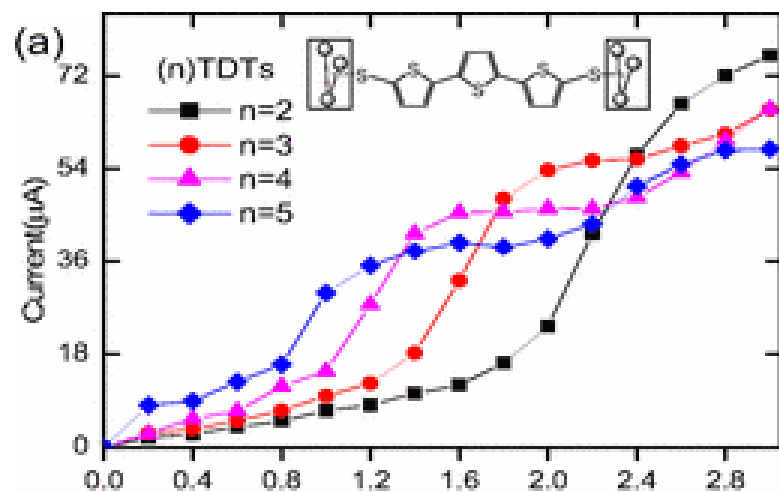


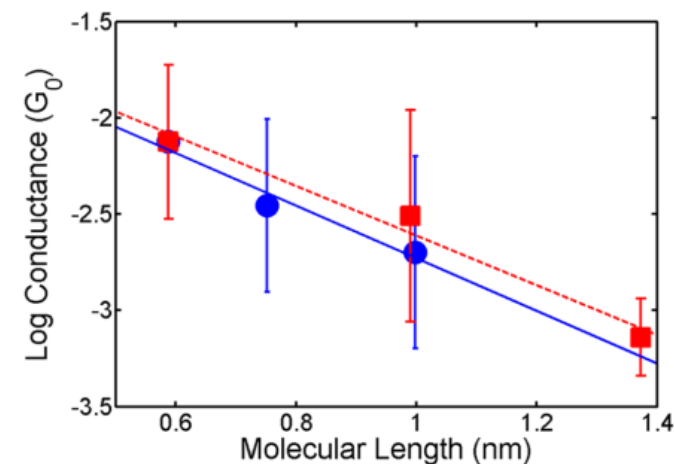
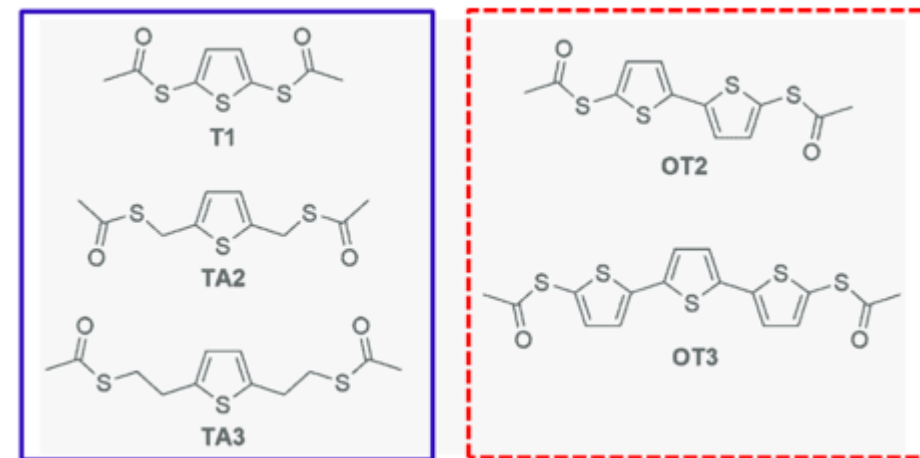
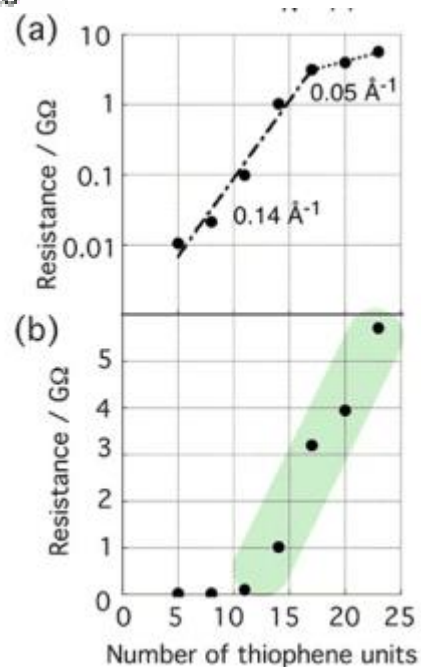
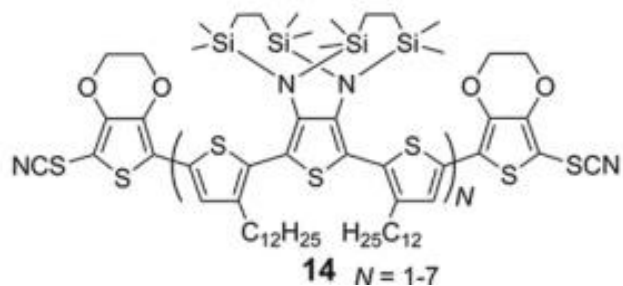
Figure 3. Typical I - V curves measured on molecular junctions. $T = 4.2 \text{ K}$.

DOI: 10.1103/PhysRevLett.88.226801

Why Oligothiophene?

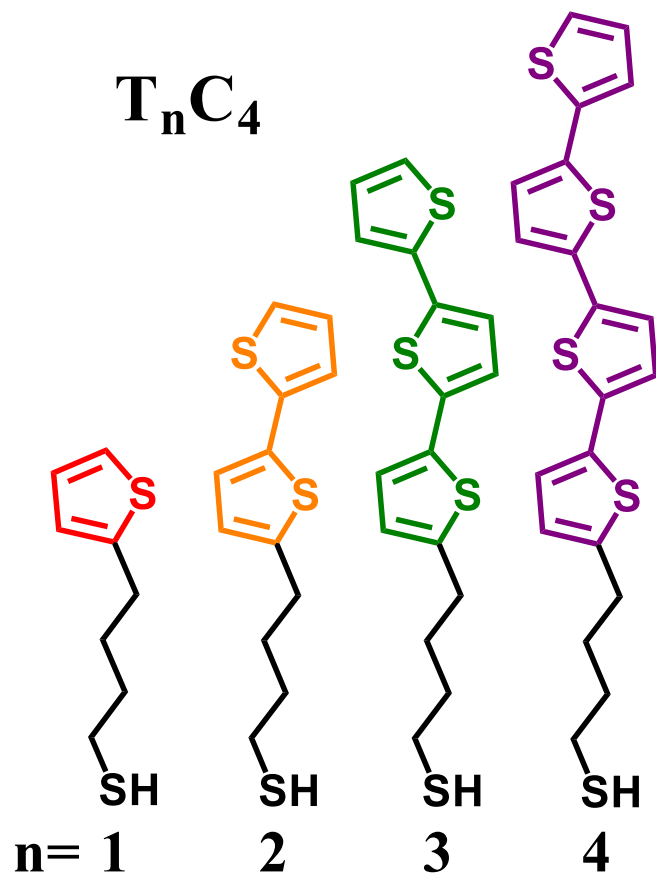



Phys. Rev. B 75, 245407



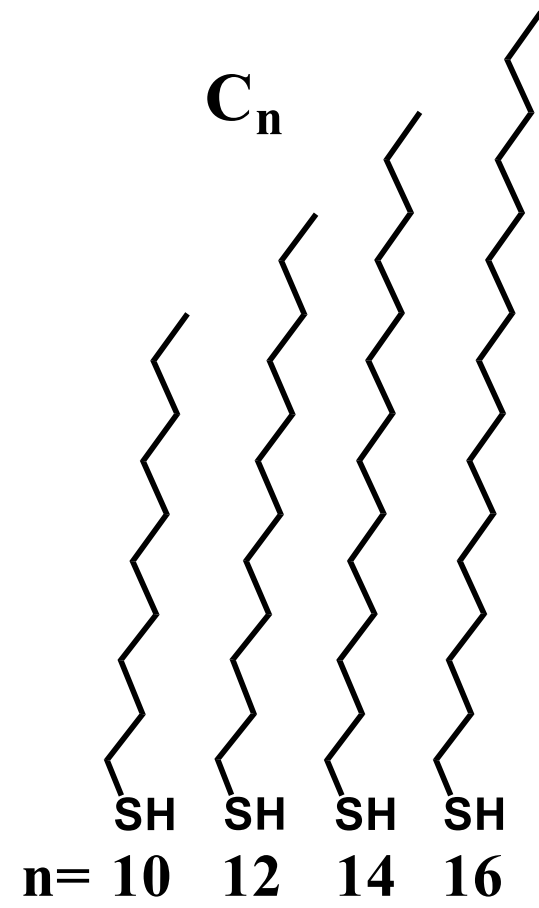
[dx.doi.org/10.1021/cm504254n](https://doi.org/10.1021/cm504254n) | Chem. Mater. 2014, 26, 7229–7235


Target system



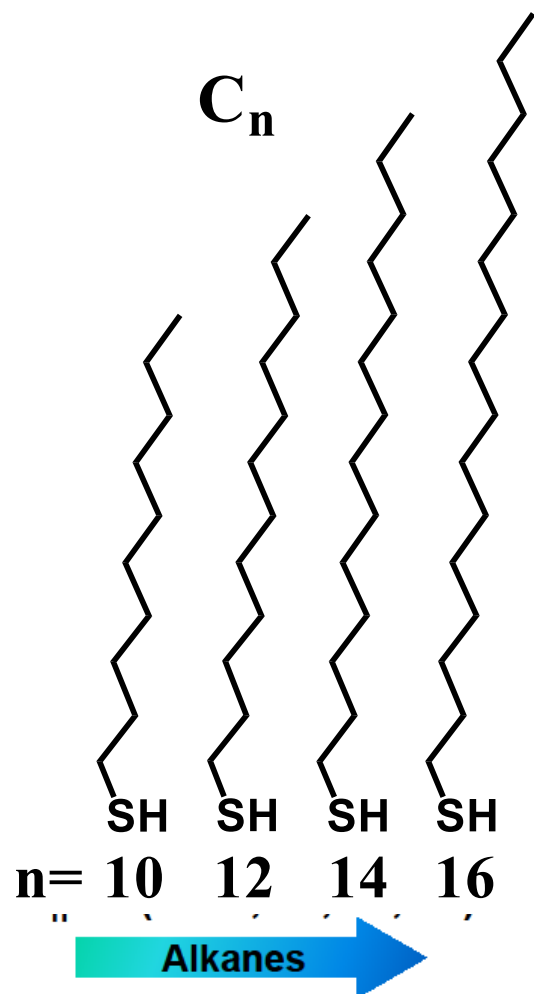
Oligothiophenes 
 Increase in Tunneling Distance (d)
 Decrease in Barrier Height (ϕ)

Oligothiophene series
(alkanes for reference)



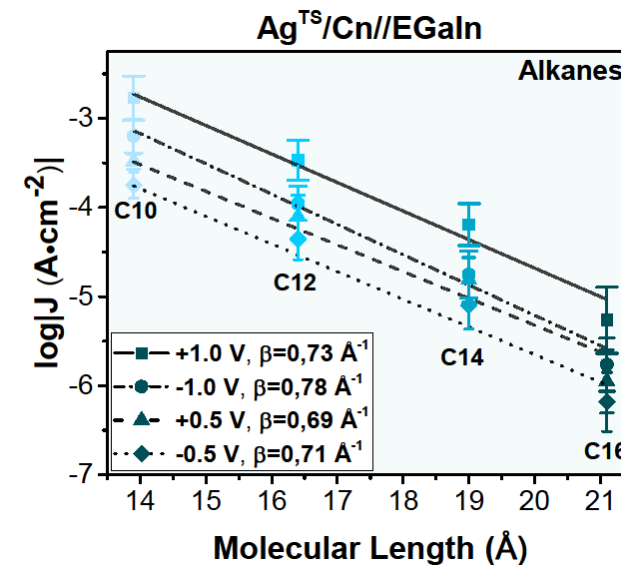
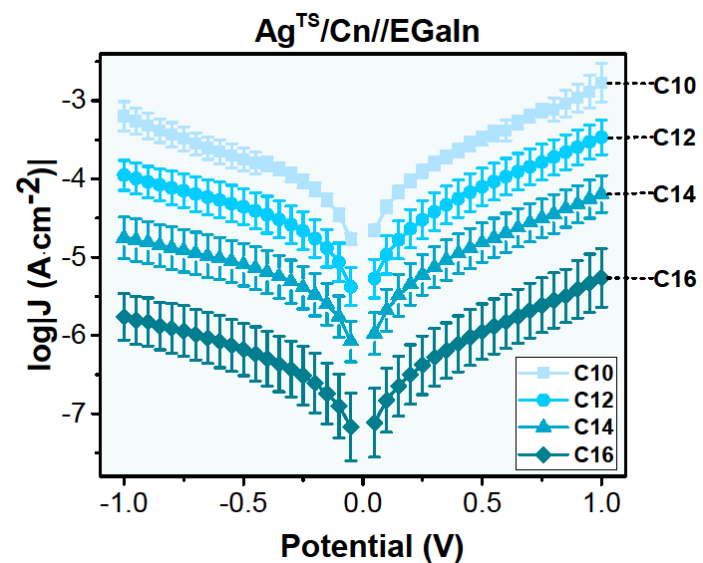
Alkanes 
 Increase in Tunneling Distance (d)
 Constant Barrier Height (ϕ)

Observation - Alkanethiols

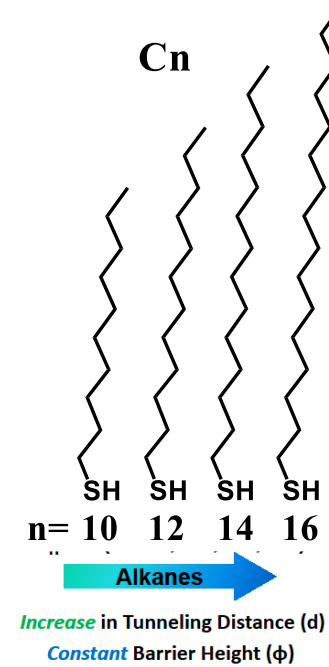
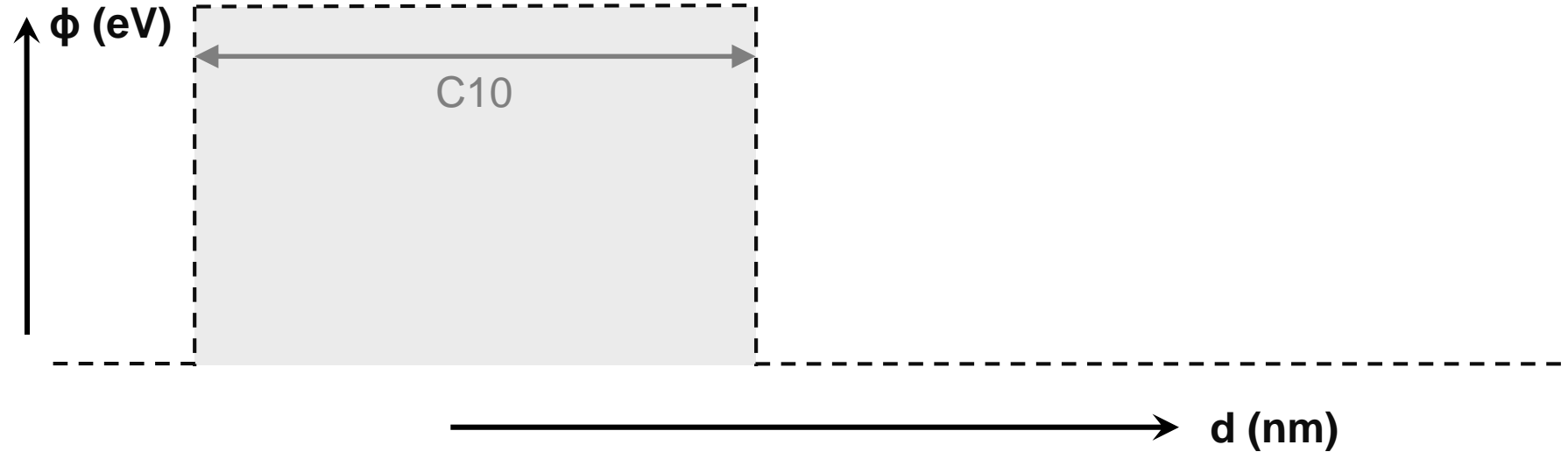


Increase in Tunneling Distance (d)

Constant Barrier Height (ϕ)



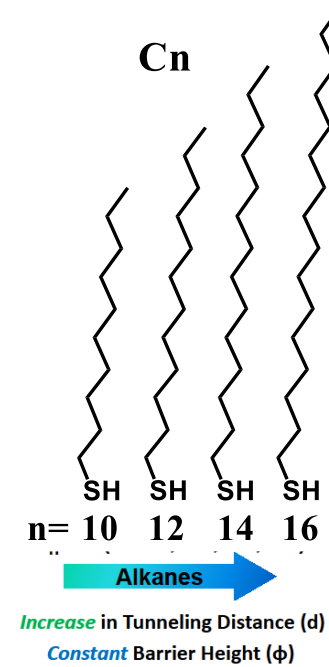
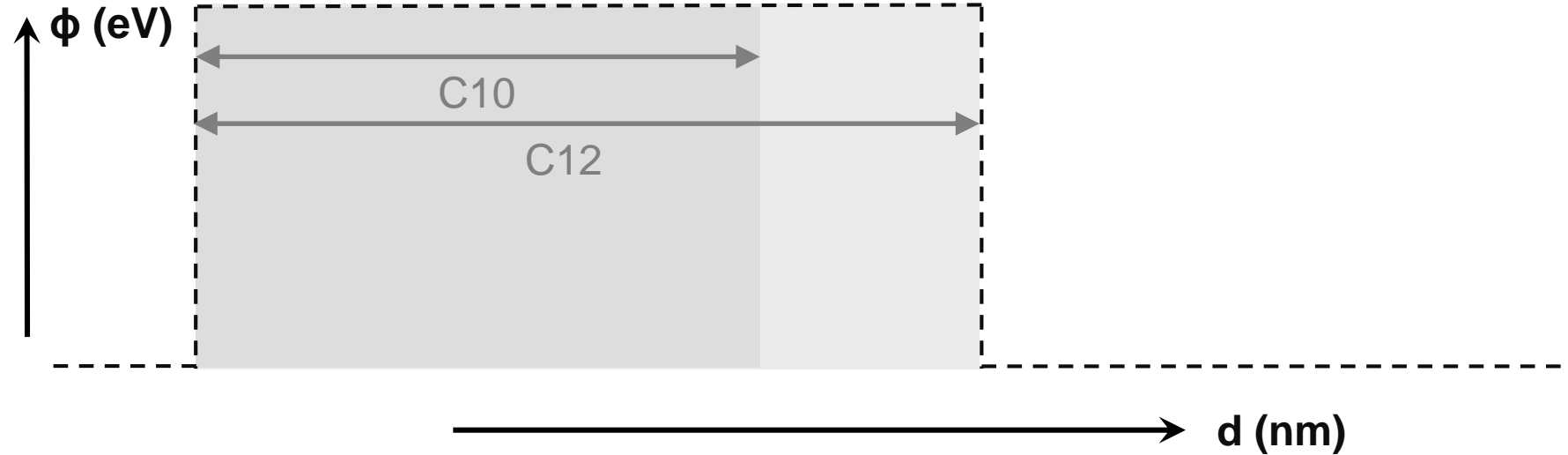
Shape of potential barriers



Single Rectangular Barrier

$$I(d) = I_0 e^{-\beta d} \quad (\beta = \text{constant})$$

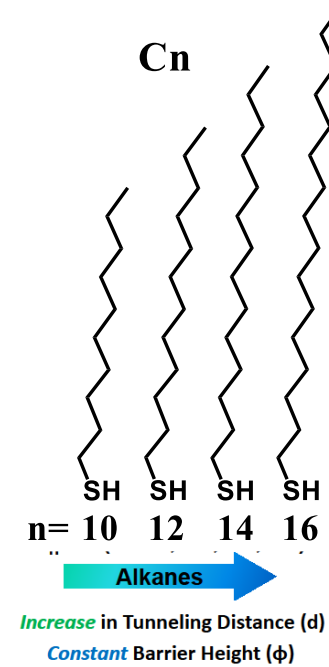
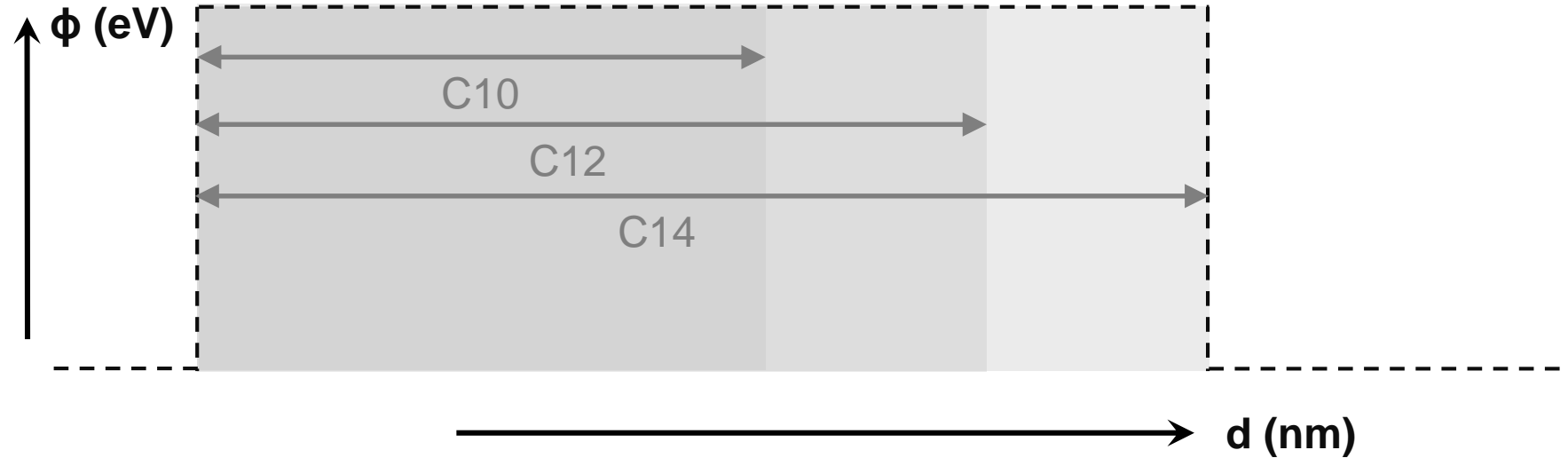
Shape of potential barriers



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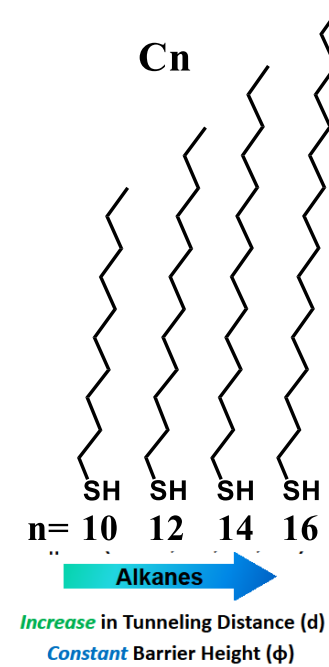
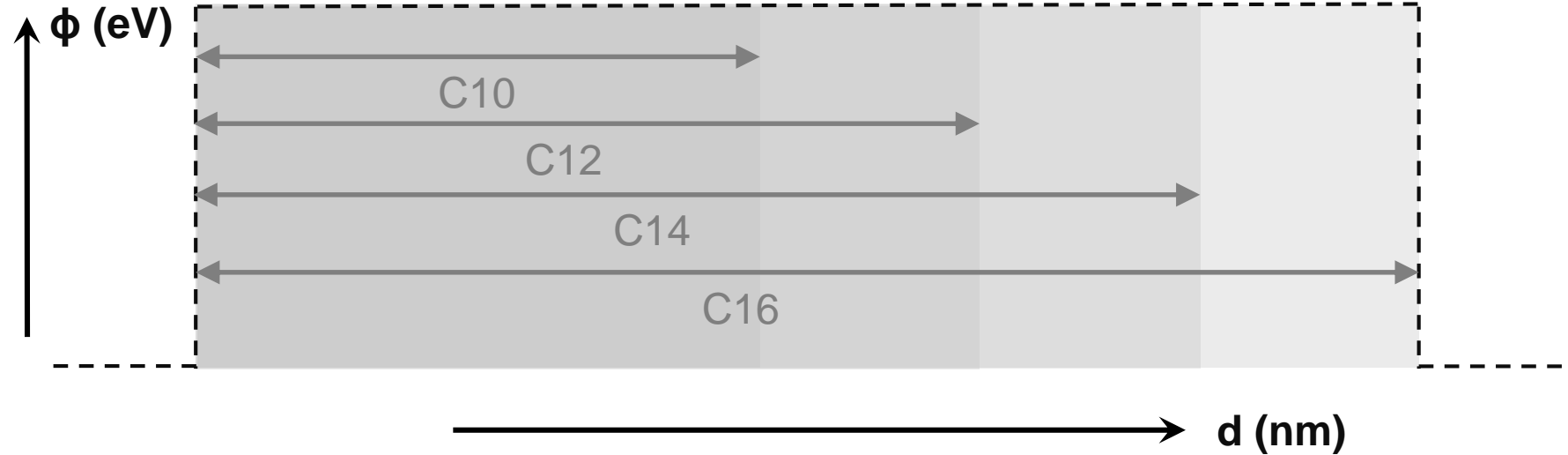
Shape of potential barriers



Single Rectangular Barrier

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Shape of potential barriers



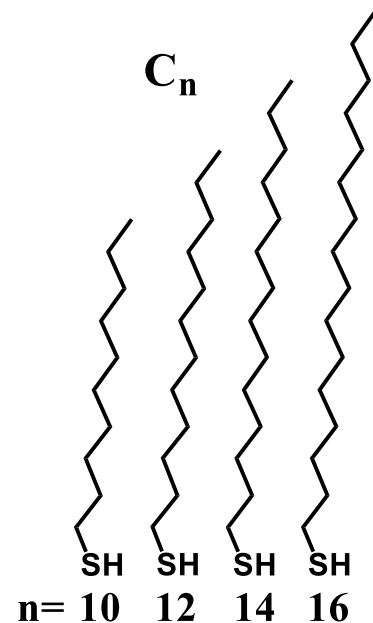
Single Rectangular Barrier

$$I(d) = I_0 e^{-\beta d}$$

(β = constant)

Observation - Alkanethiols

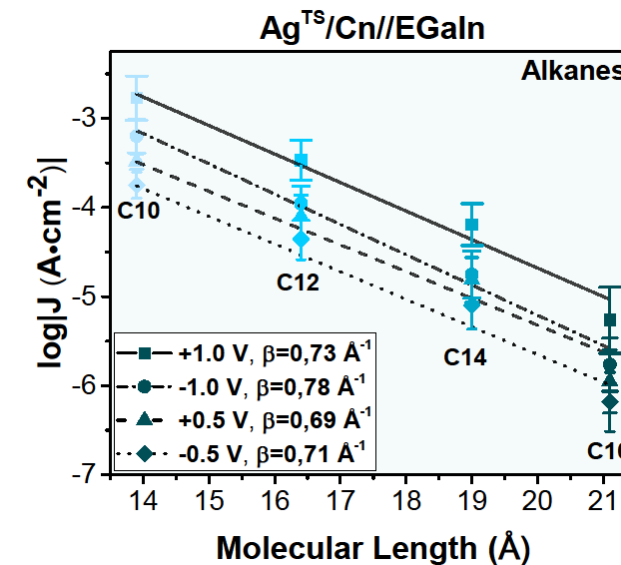
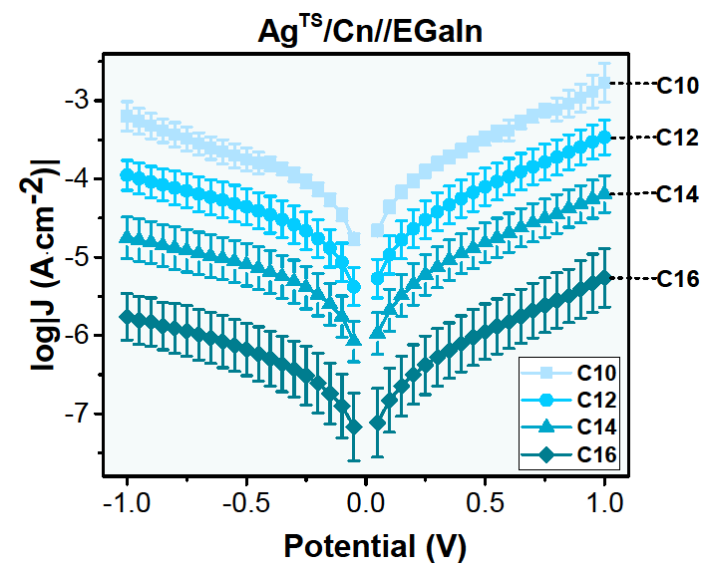
Made sense!



Alkanes

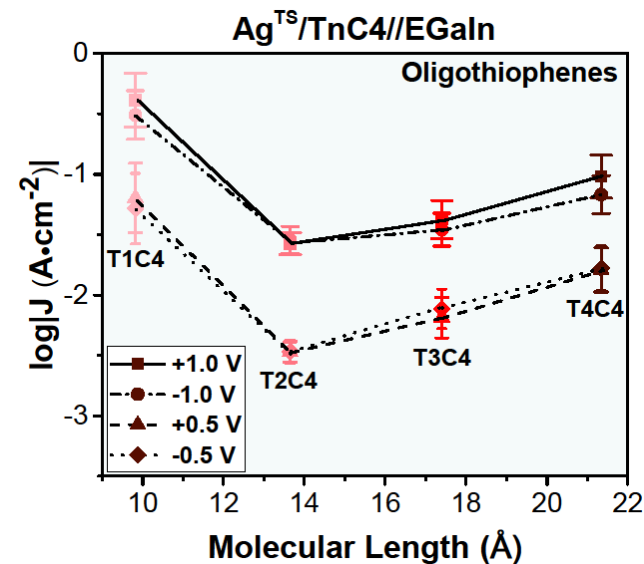
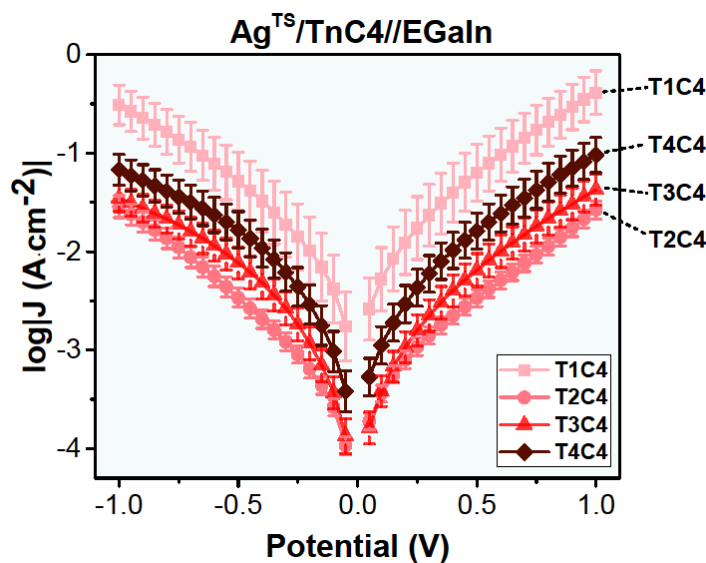
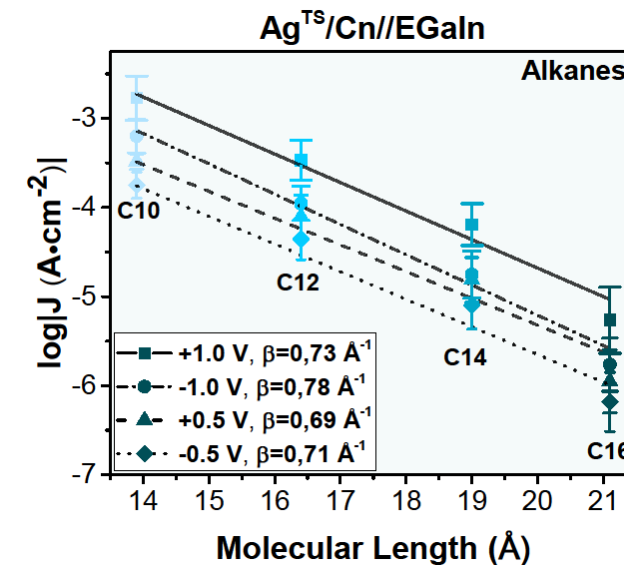
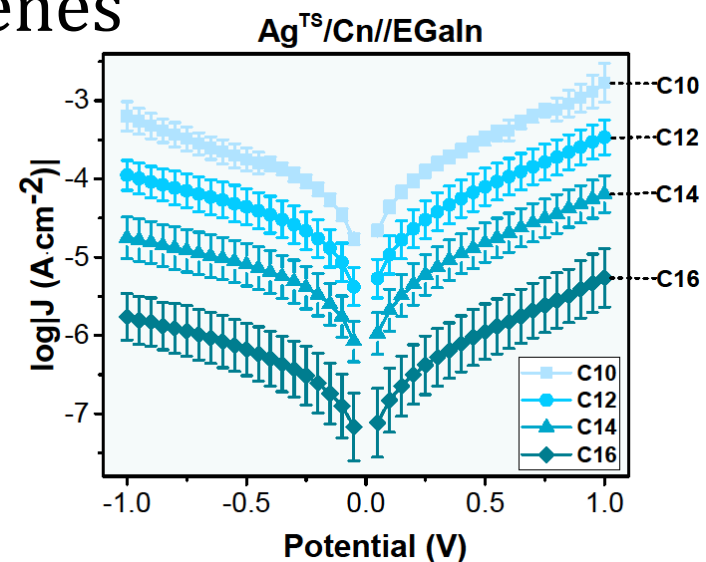
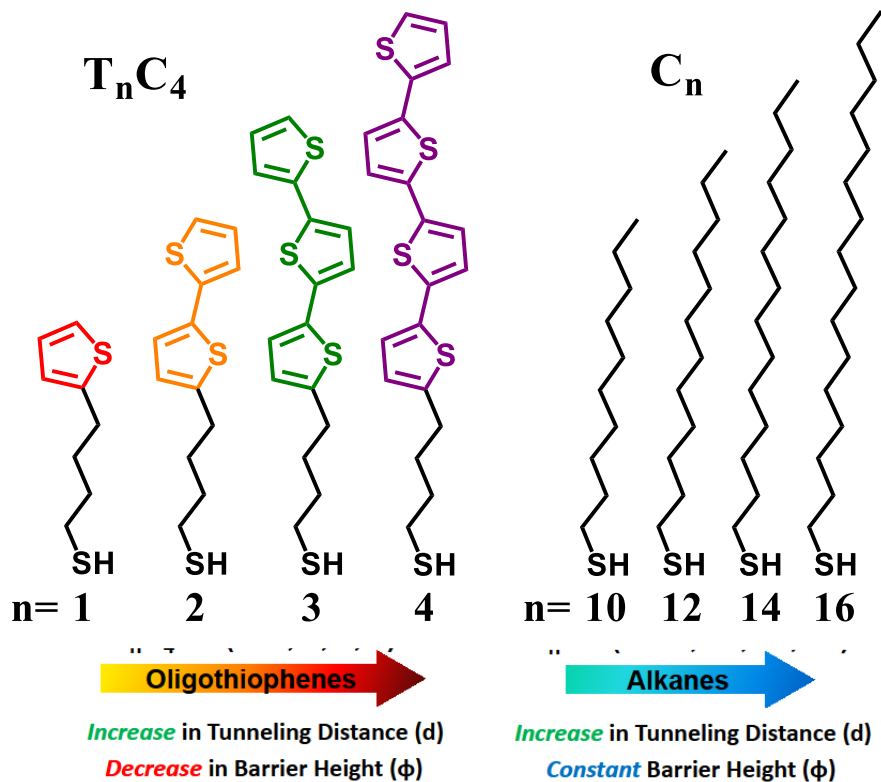
Increase in Tunneling Distance (d)

Constant Barrier Height (ϕ)



Observation - Oligothiophenes

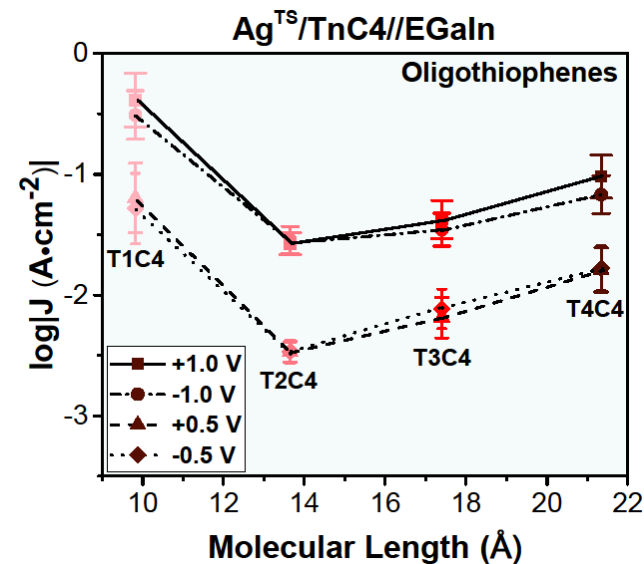
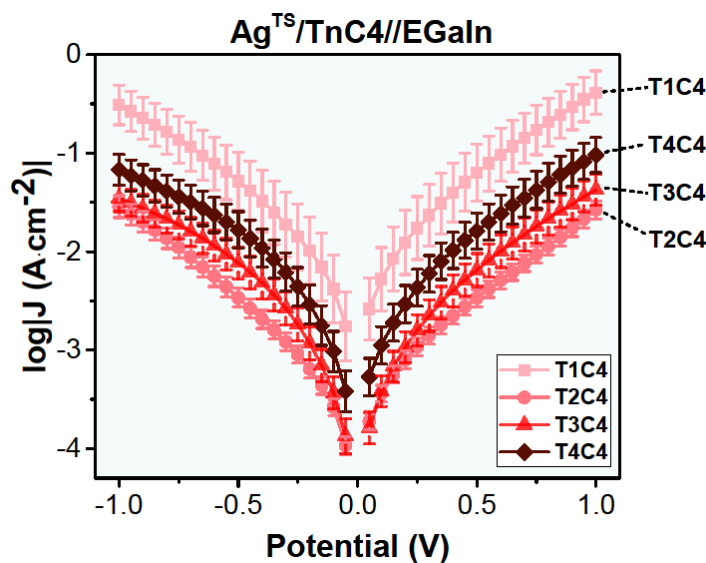
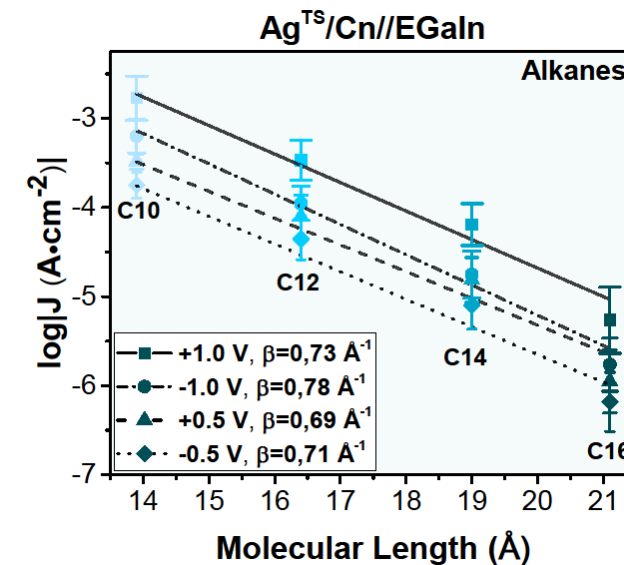
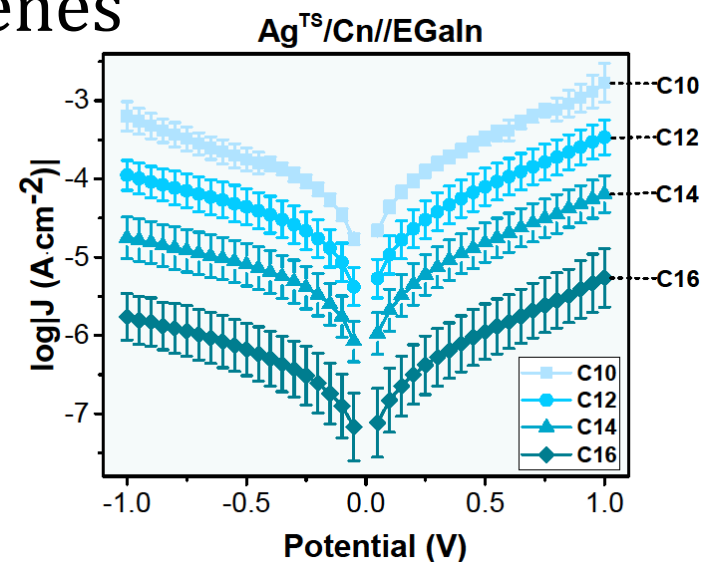
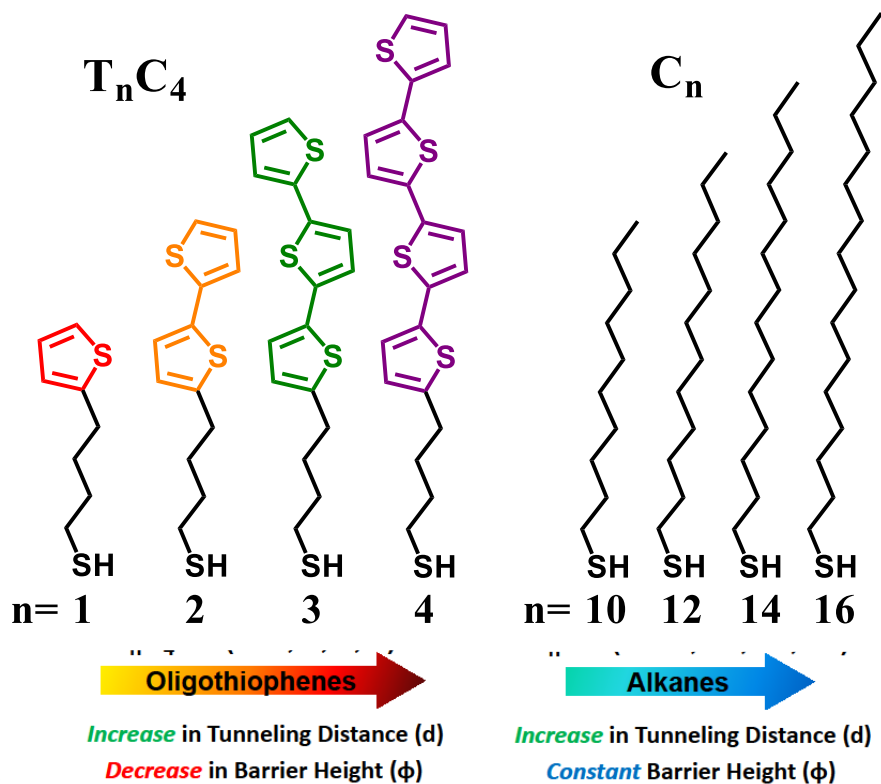
Made NO sense!



Observation - Oligothiophenes

Made NO sense!

T1C4 > **T4C4** > **T3C4** > **T2C4**

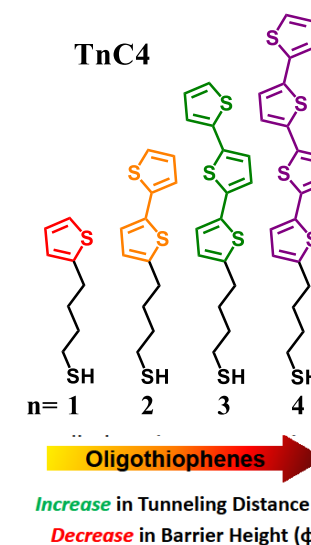
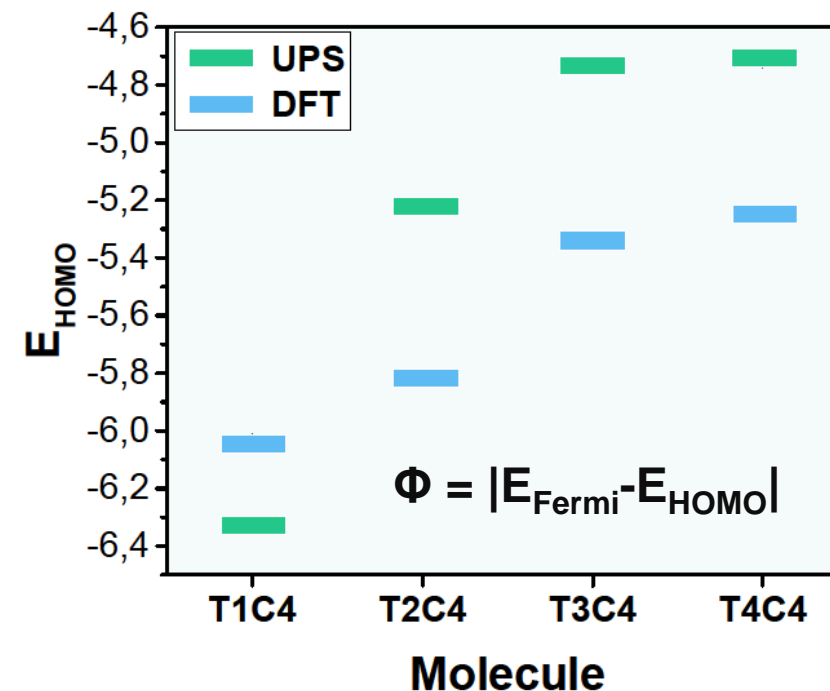
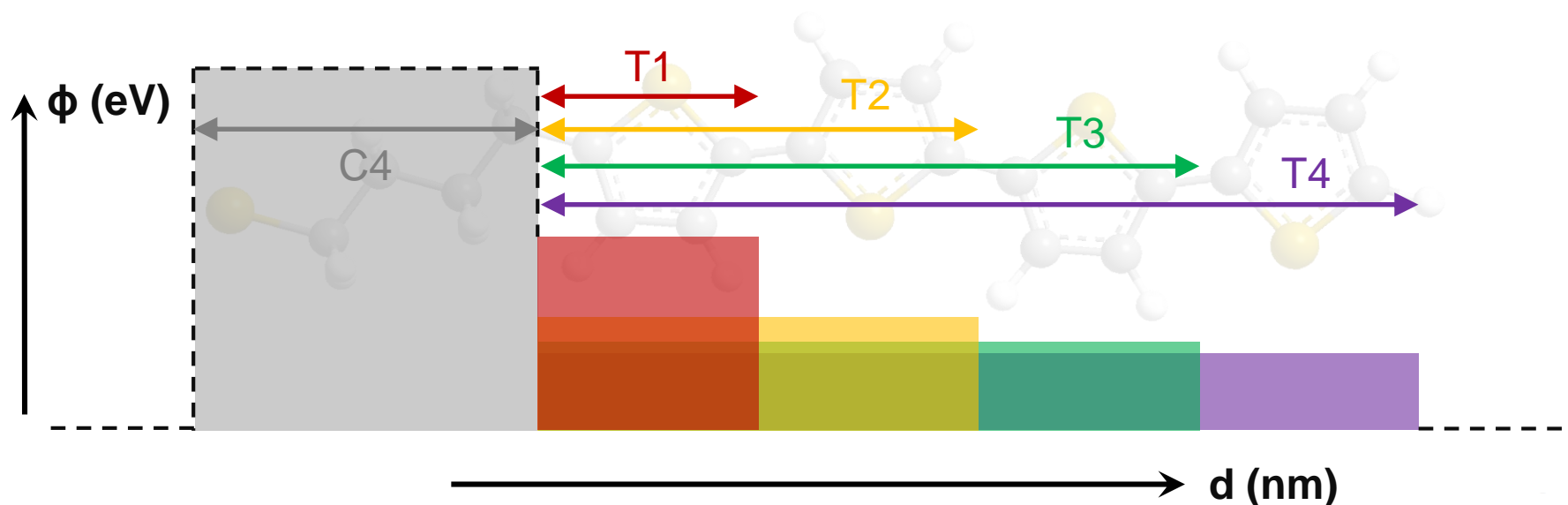


Shape of potential barriers

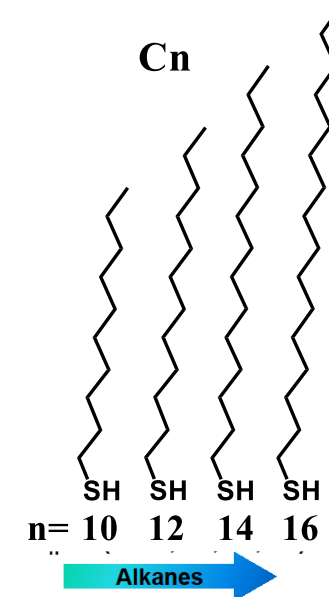
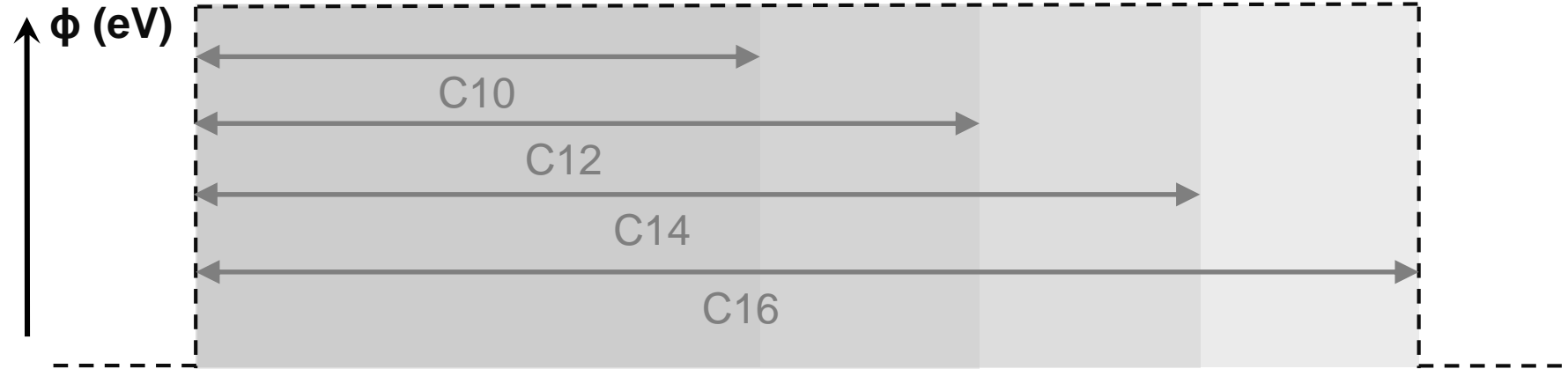
Made NO sense!

T1C4 > T4C4 > T3C4 > T2C4

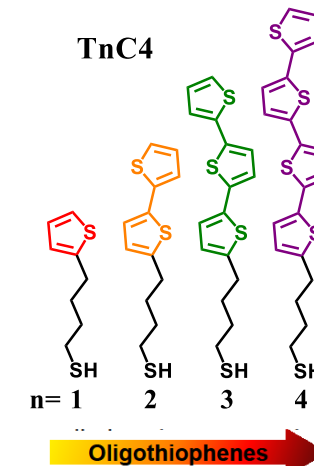
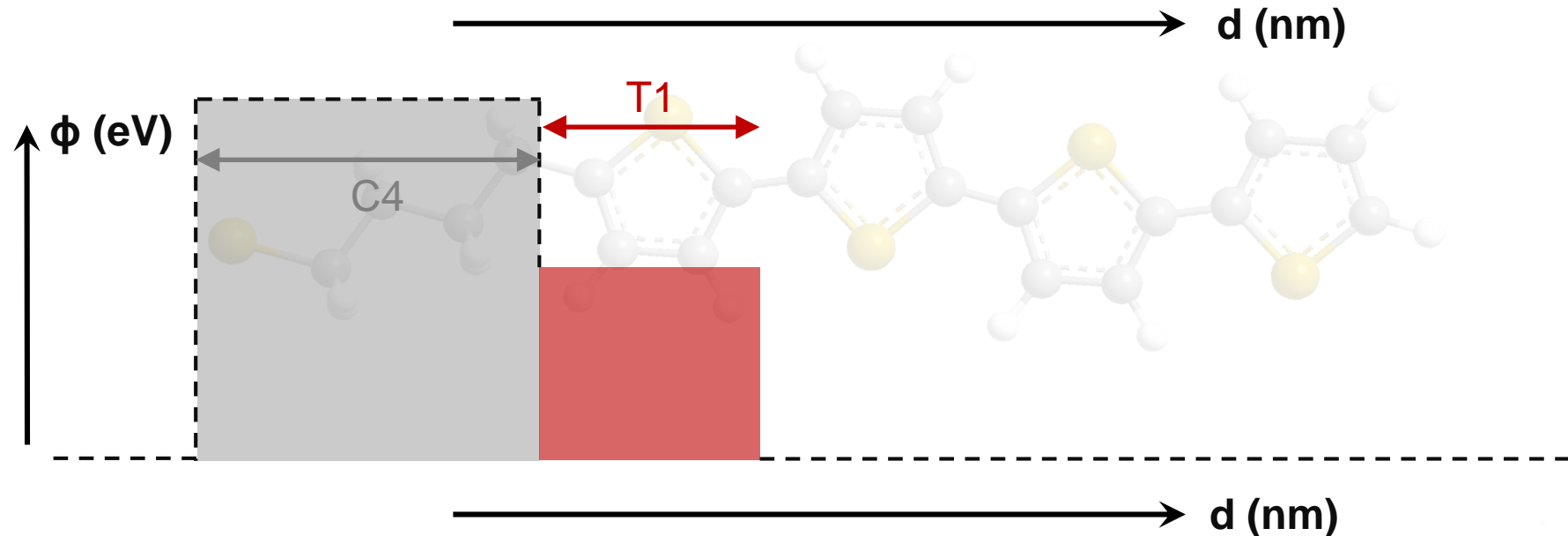
- **Two Barrier Model** taking into account:
 1. increasing barrier width
 2. decreasing barrier height



Shape of potential barriers

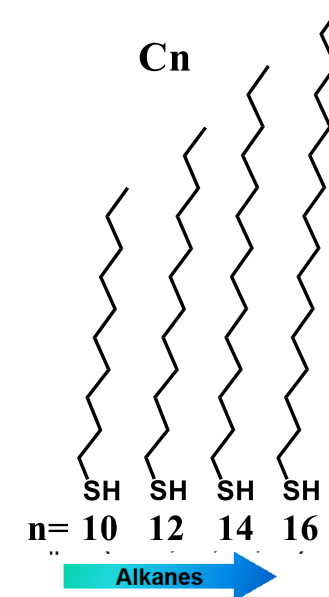
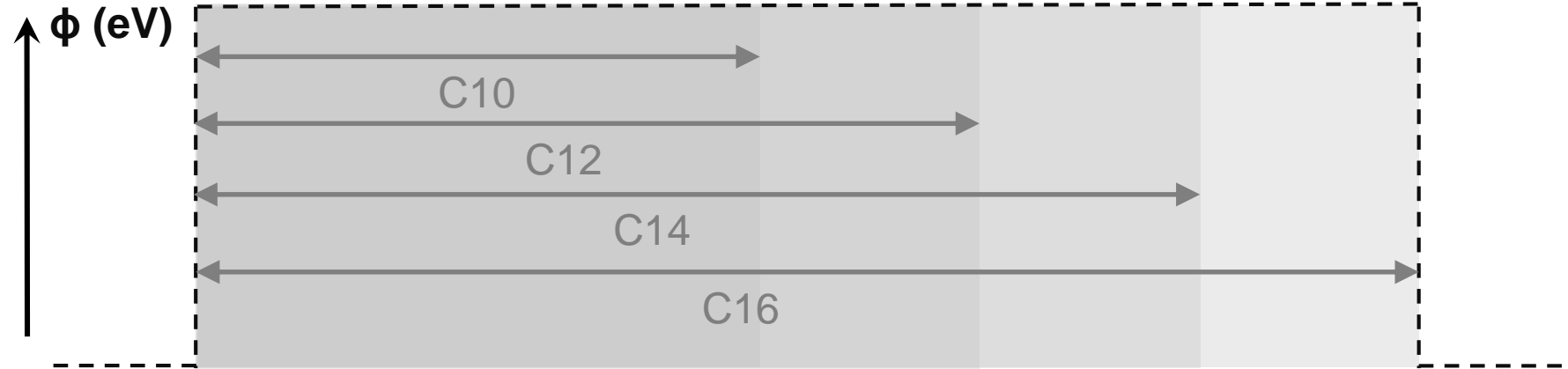


Increase in Tunneling Distance (d)
 Constant Barrier Height (ϕ)

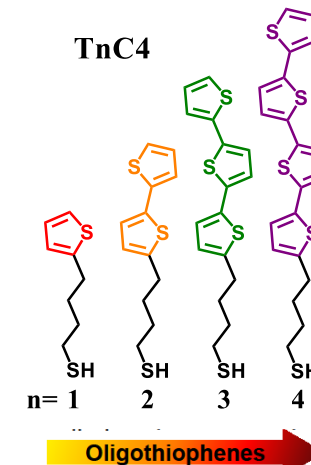
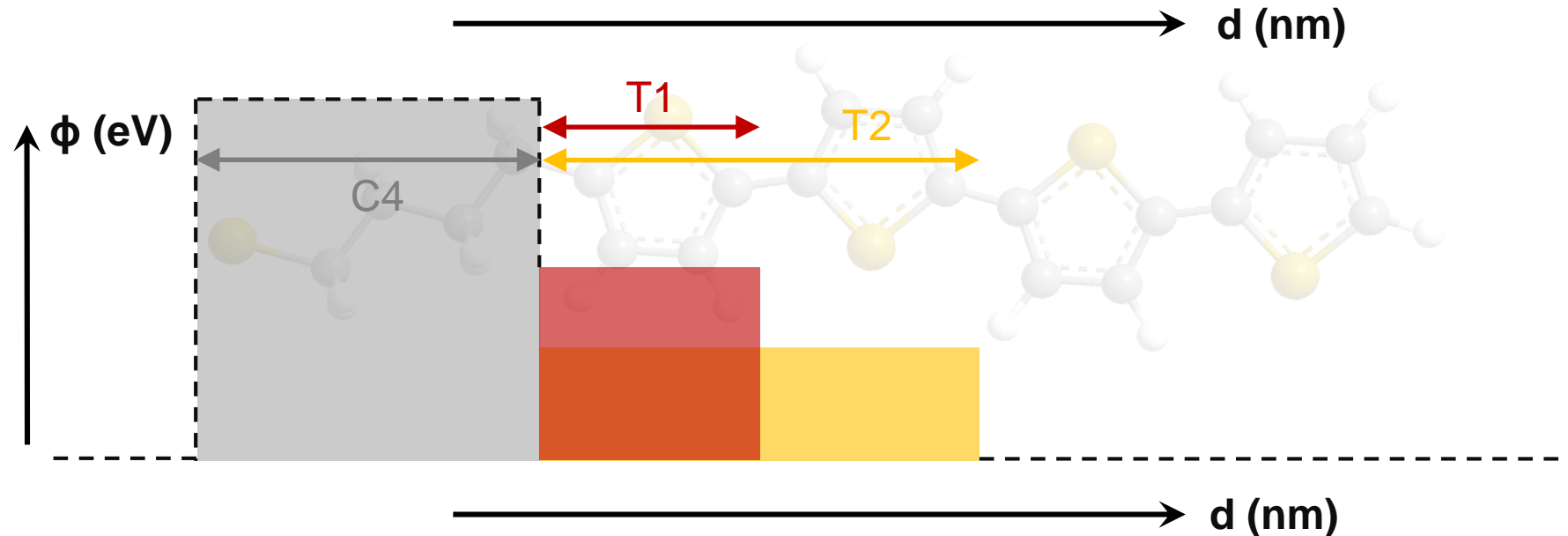


Increase in Tunneling Distance (d)
 Decrease in Barrier Height (ϕ)

Shape of potential barriers

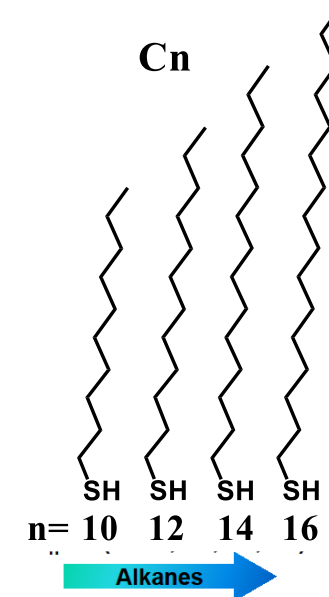
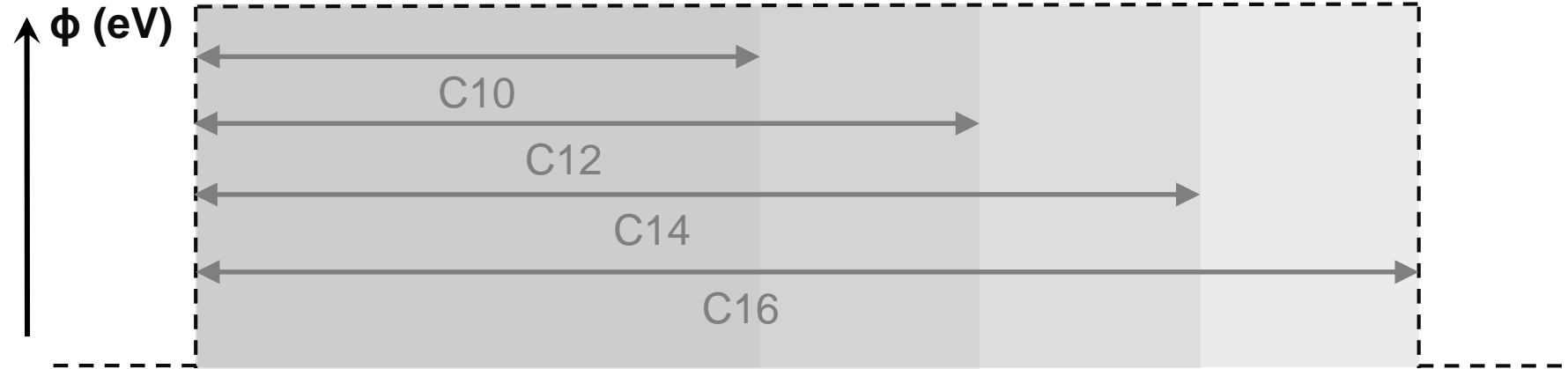


Increase in Tunneling Distance (d)
 Constant Barrier Height (ϕ)

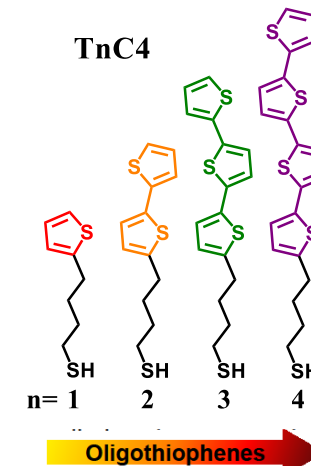
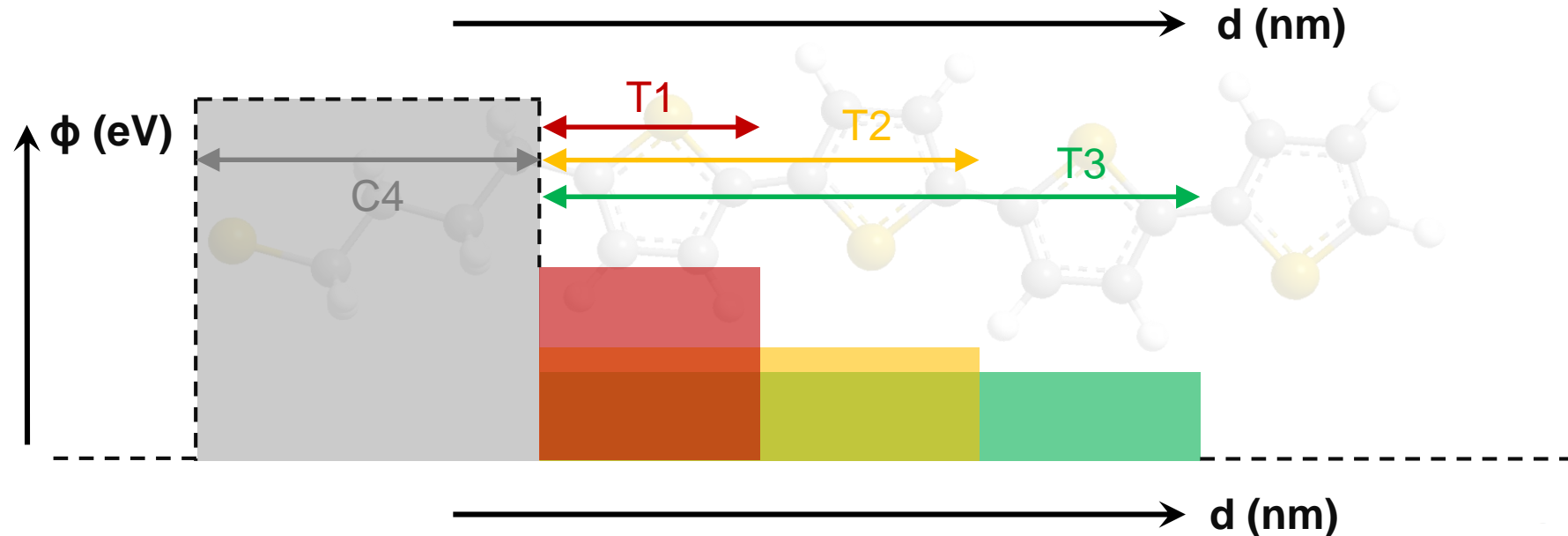


Increase in Tunneling Distance (d)
 Decrease in Barrier Height (ϕ)

Shape of potential barriers

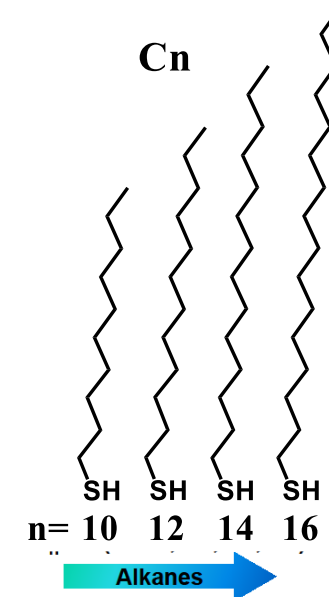
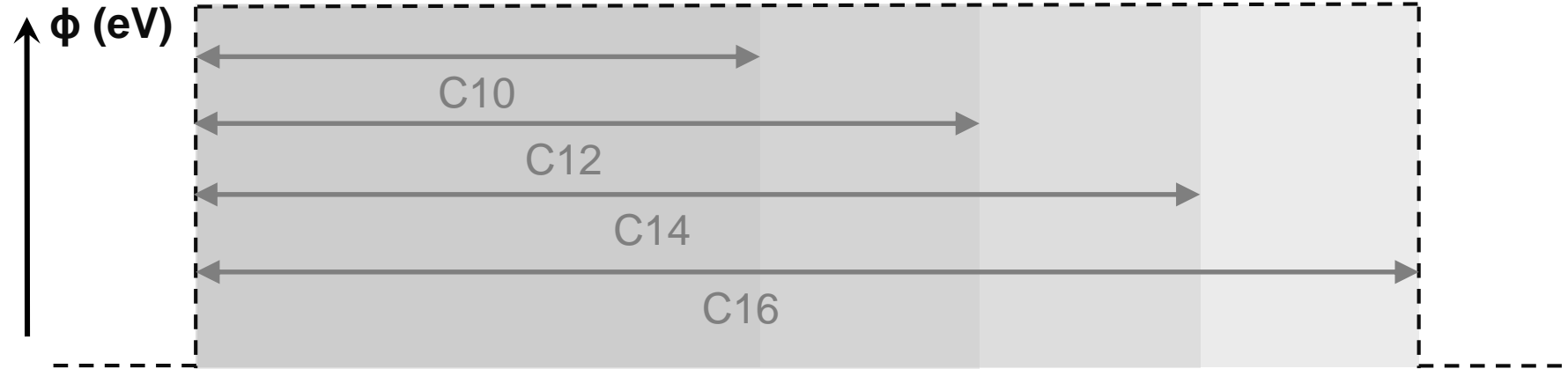


Increase in Tunneling Distance (d)
 Constant Barrier Height (ϕ)

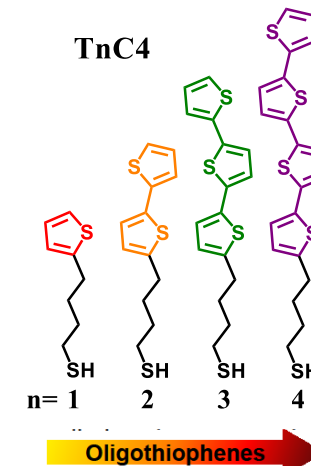
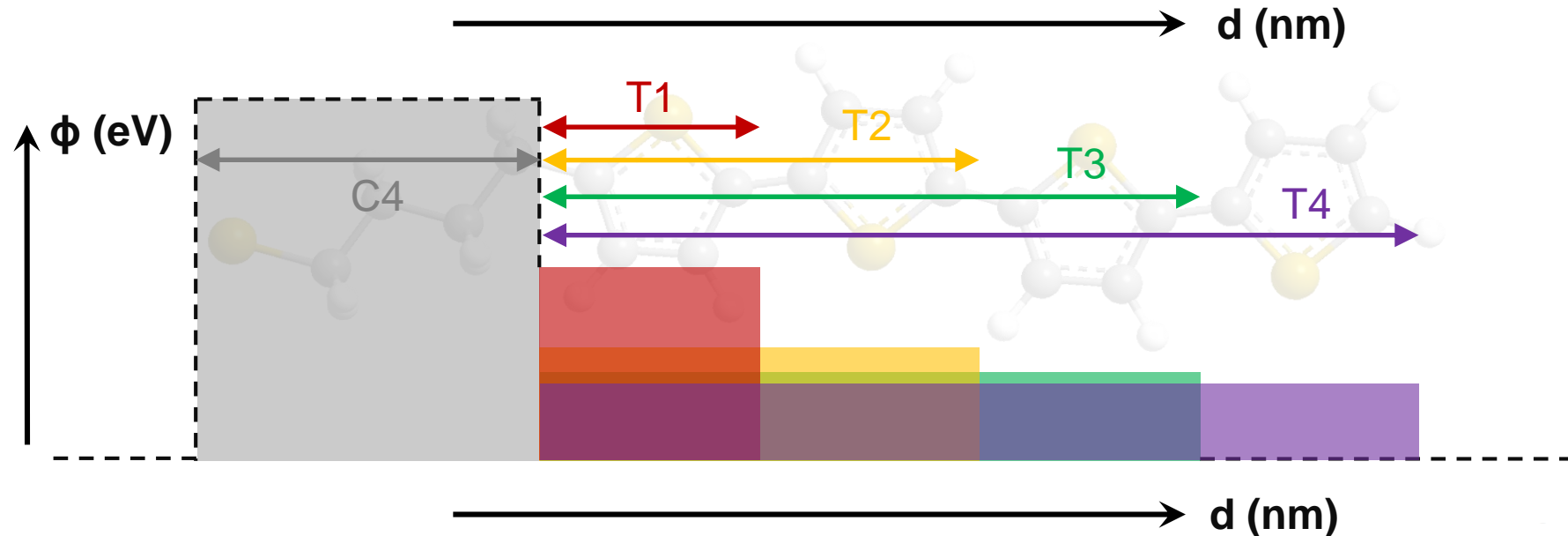


Increase in Tunneling Distance (d)
 Decrease in Barrier Height (ϕ)

Shape of potential barriers



Increase in Tunneling Distance (d)
 Constant Barrier Height (ϕ)

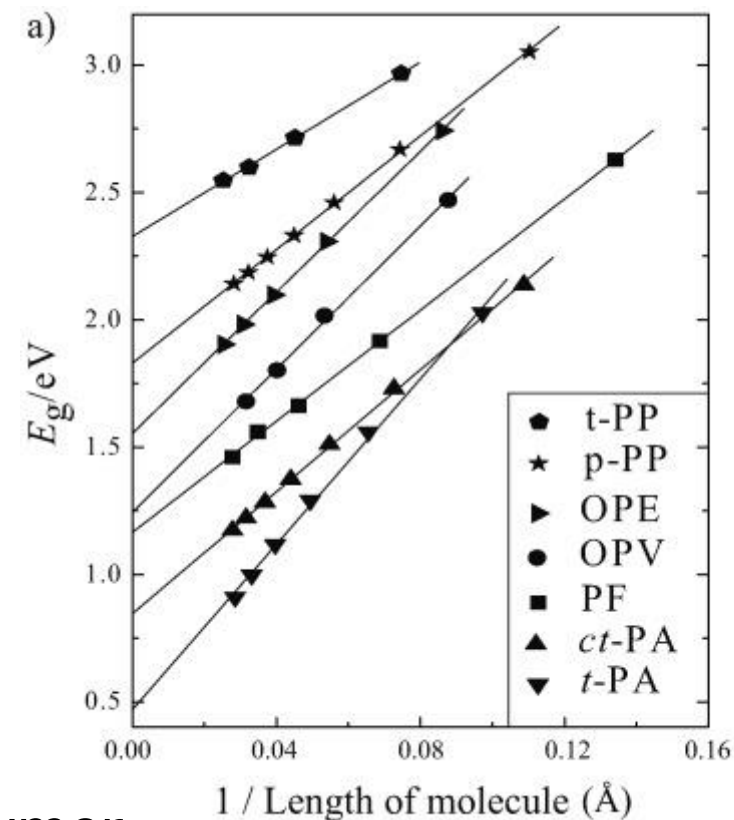


Increase in Tunneling Distance (d)
 Decrease in Barrier Height (ϕ)

Known dependence of bandgap on molecular length

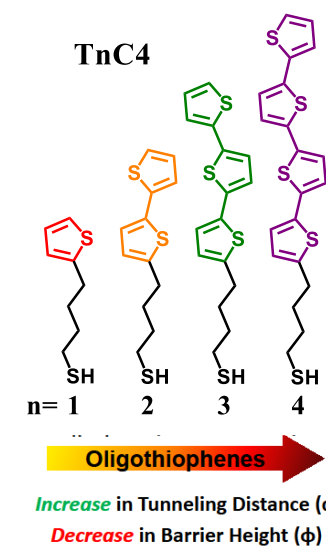
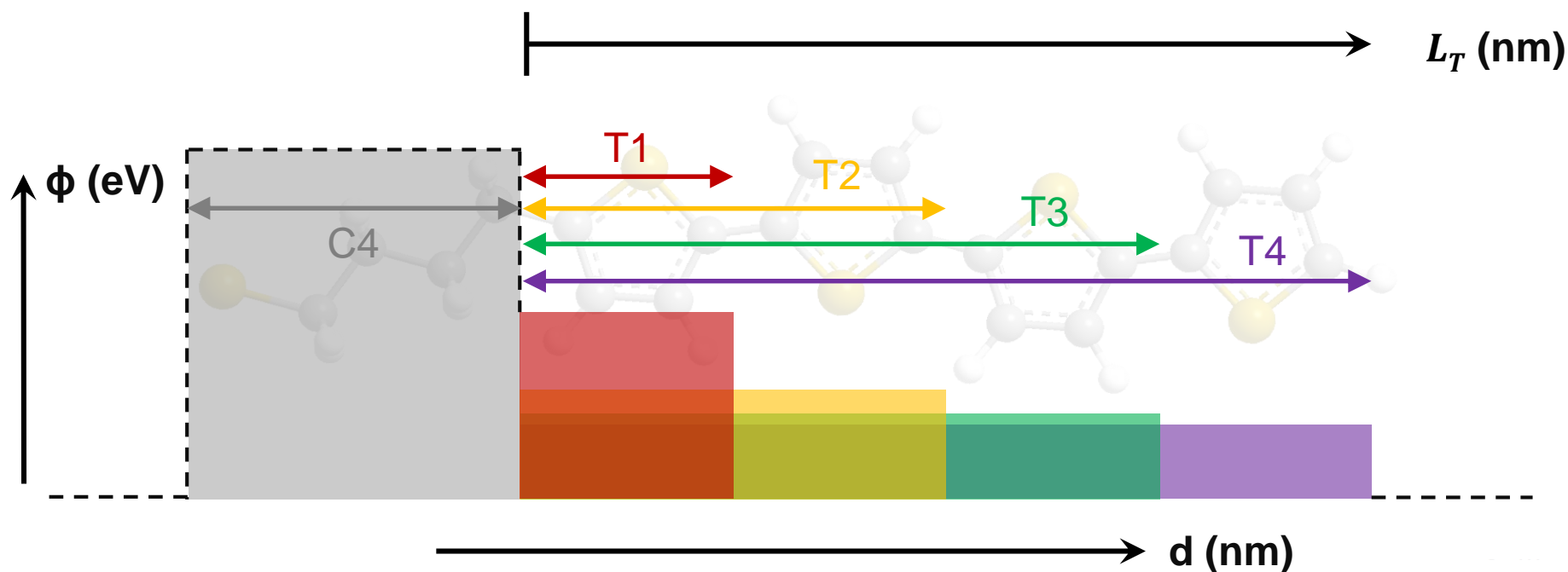
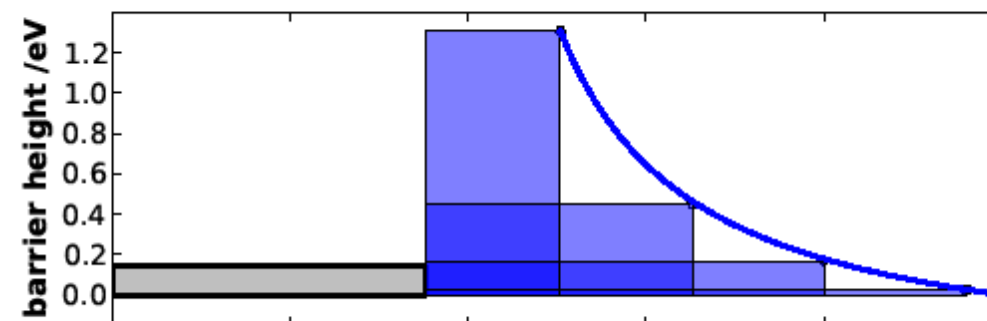
$$E_g(d) = \Delta^\infty + \frac{\alpha}{d}$$

> Δ^∞ would be the value of bandgap of an infinite polymer



Shape of potential barriers

$$\phi(L_T) = \Delta^\infty + \frac{\alpha}{L_T}$$



Solving 2-Barrier Model

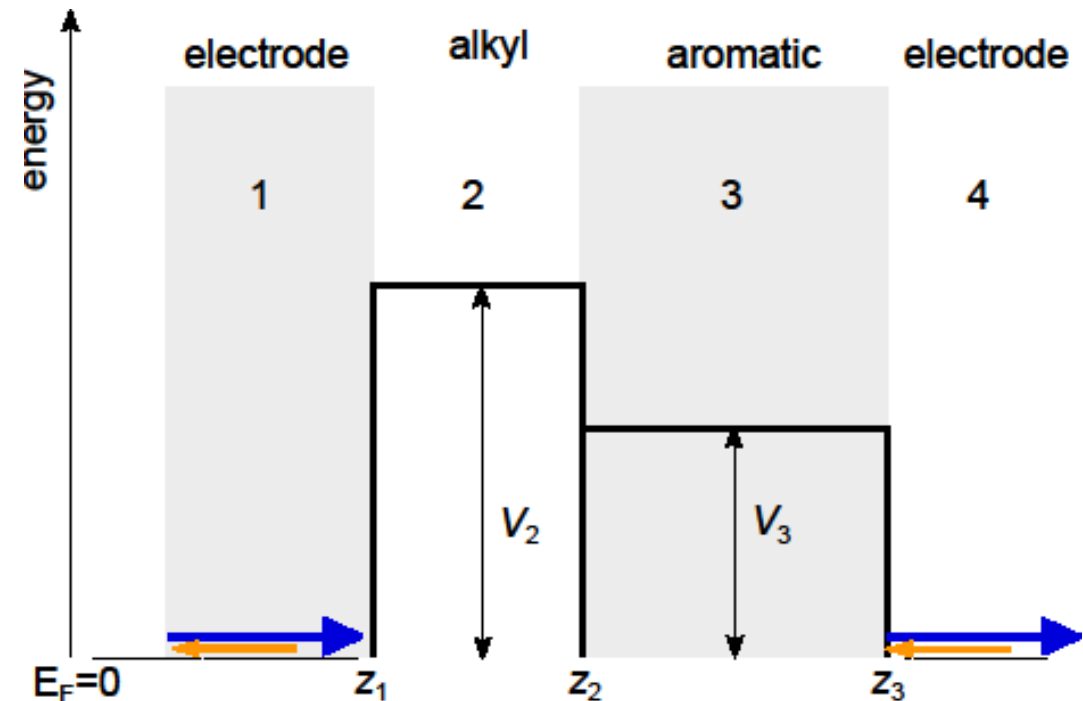
$$\psi_p(z) = A_p e^{ik_p z} + B_p e^{-ik_p z} \quad (z_{p-1} < z < z_p). \quad k_p = \sqrt{\frac{2me}{\hbar^2} (E - V_p)}$$

$$\psi_p(z_p) = \psi_{p+1}(z_p), \quad \text{and} \quad \frac{d\psi_p(z_p)}{dz} = \frac{d\psi_{p+1}(z_p)}{dz}.$$

$$V_3(L_T) = \Delta^\infty + \frac{\alpha}{L_T}$$

Calculating transmission:

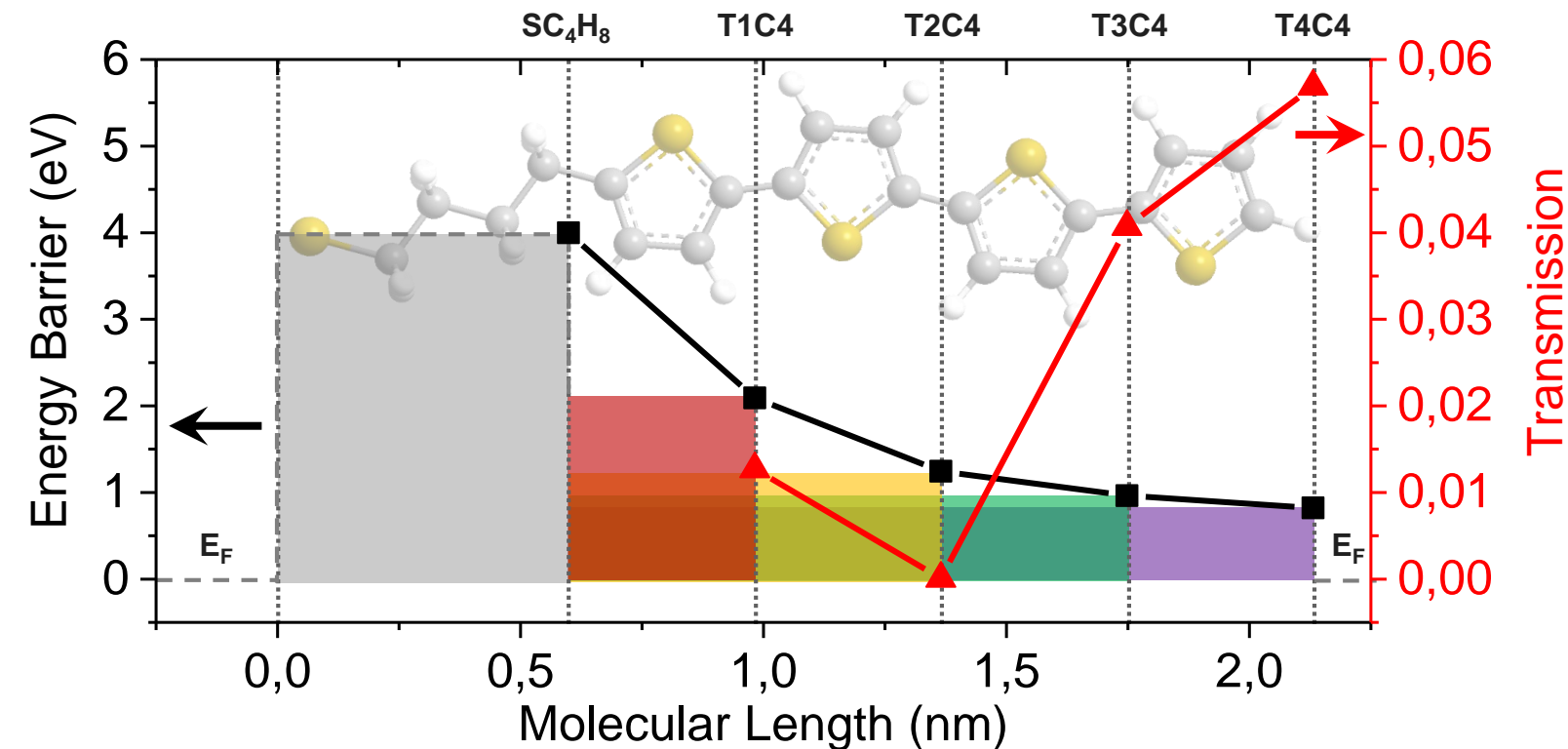
$$T(E) = \frac{J_{4 \text{ trans}}}{J_1} = \frac{|A_4|^2}{|A_1|^2}$$



Calculating transmission:

$$V_3(L_T) = \Delta^\infty + \frac{\alpha}{L_T}$$

$$T(E) = \frac{J_{4trans}}{J_1} = \frac{|A_4|^2}{|A_1|^2}$$



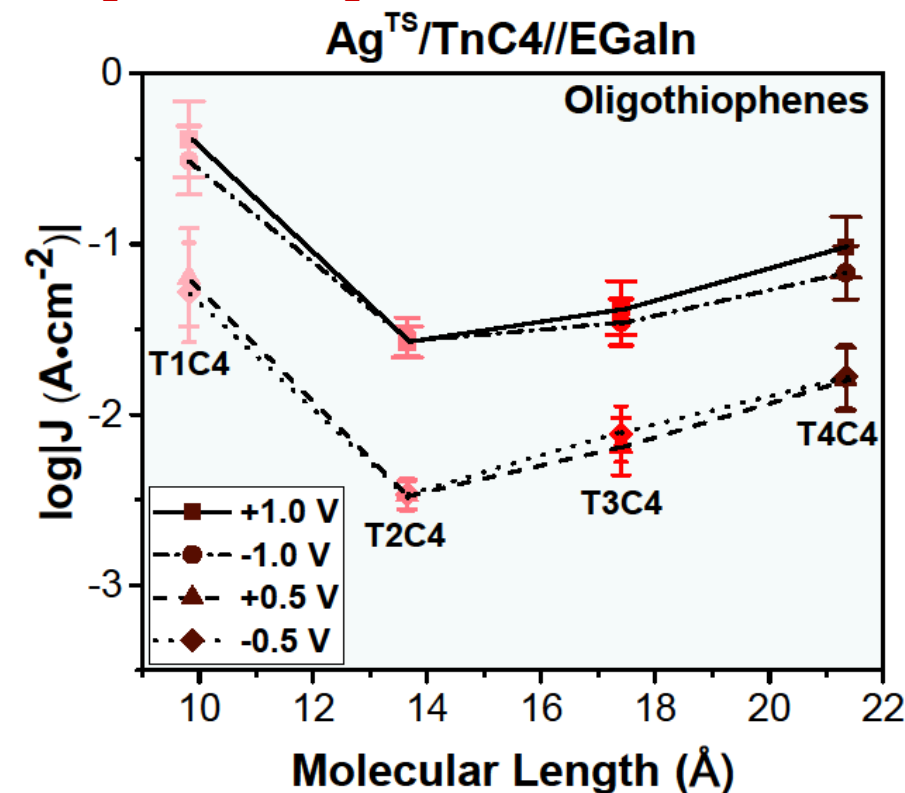
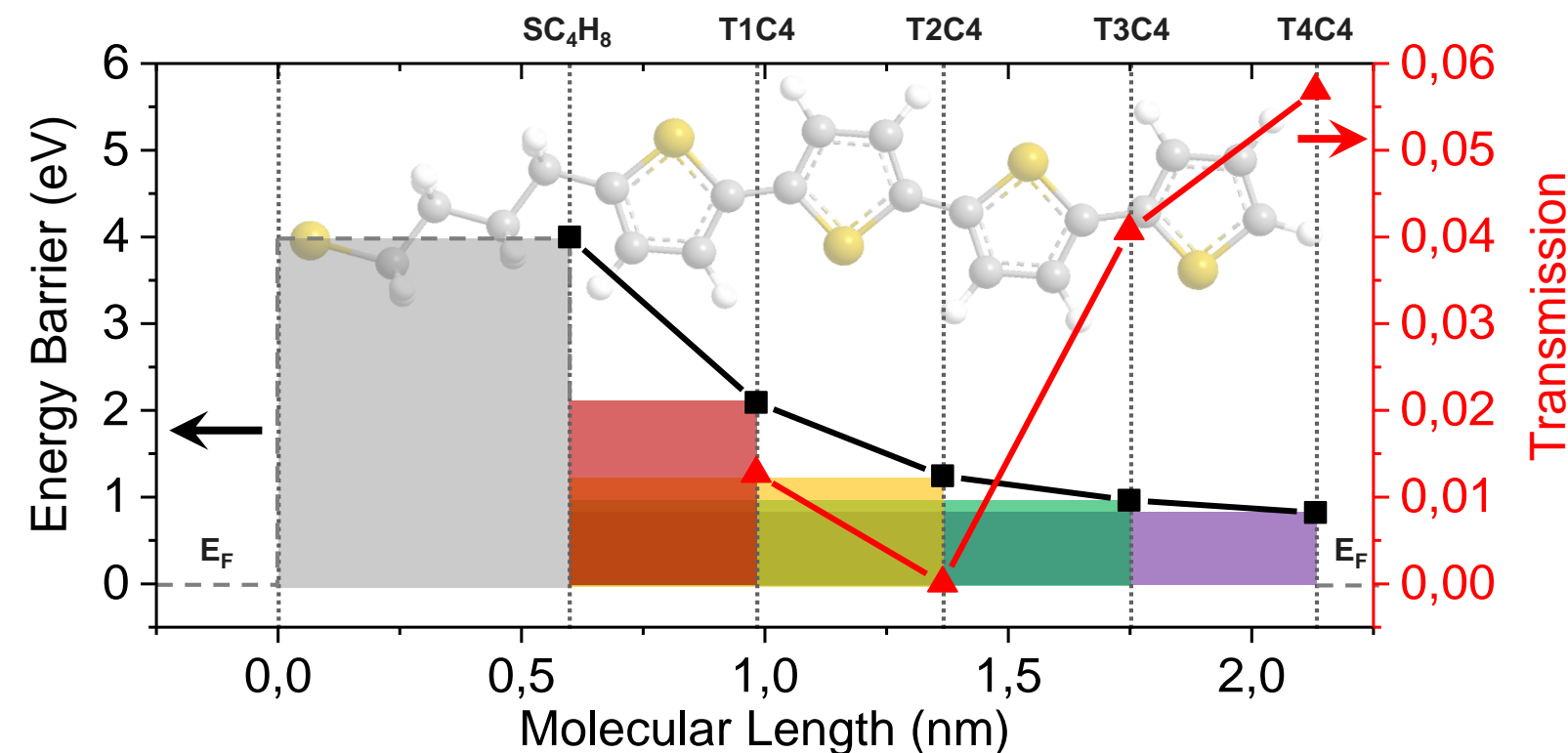
Solving 2-Barrier Model

Calculating transmission:

> The exact curvature and the trend depends on the values of parameters Δ^∞ and α .

$$V_3(L_T) = \Delta^\infty + \frac{\alpha}{L_T}$$

$$T(E) = \frac{J_{4trans}}{J_1} = \frac{|A_4|^2}{|A_1|^2}$$

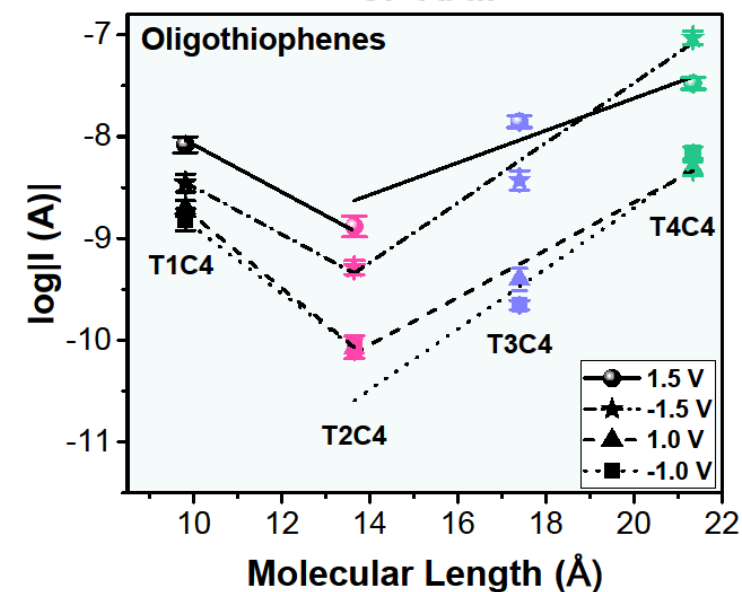
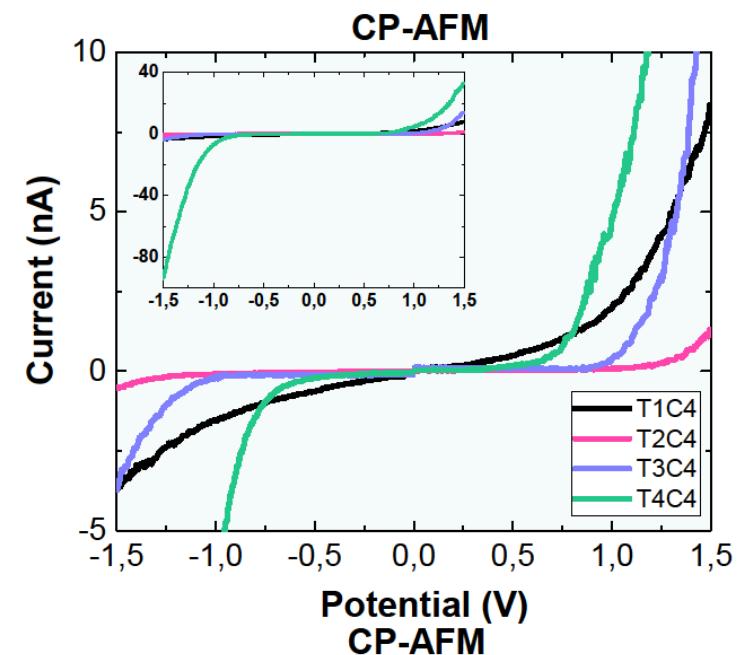
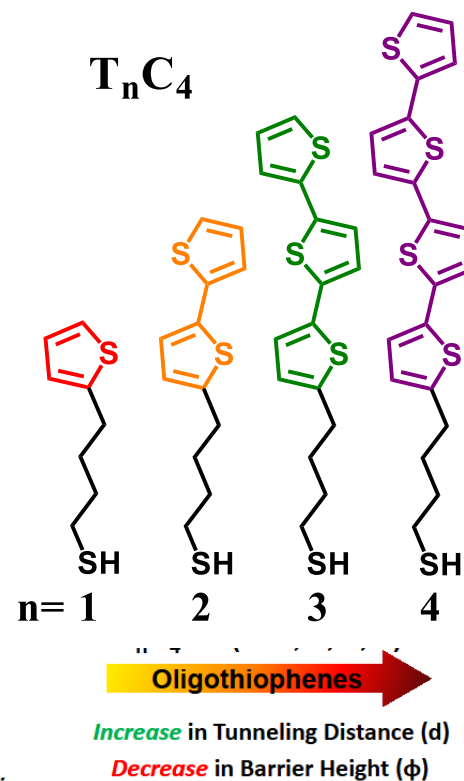


I-V and NDC Simulations

1) Simulating CP AFM I - V curves (DFT)

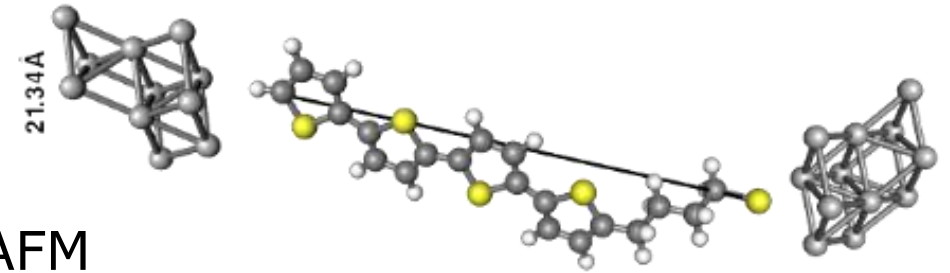
Experimental....

$T_4C_4 > (T_3C_4 \sim T_1C_4) > T_2C_4$



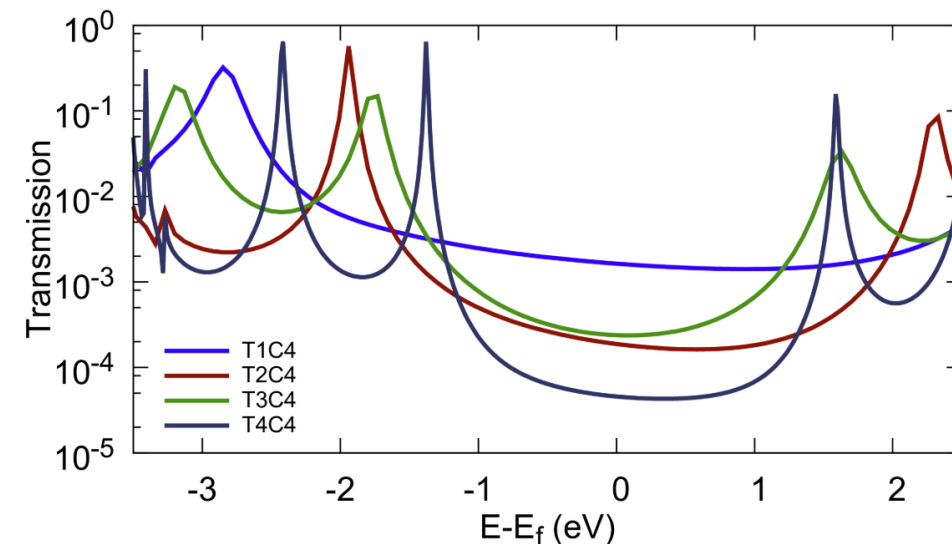
1) Simulating CP AFM I - V curves (DFT)

- › Single-molecular DFT calculations closer to CP AFM
- › Input: $T(E)$ from DFT on single-molecular junctions



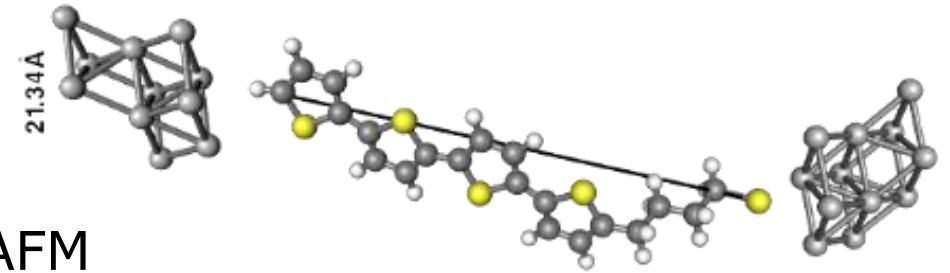
$$I(V) = \frac{2e}{h} c \int_{E_F - eV/2}^{E_F + eV/2} T(E) \cdot dE$$

(Chem. Soc. Rev.2015,44, 875–888)



1) Simulating CP AFM I - V curves (DFT)

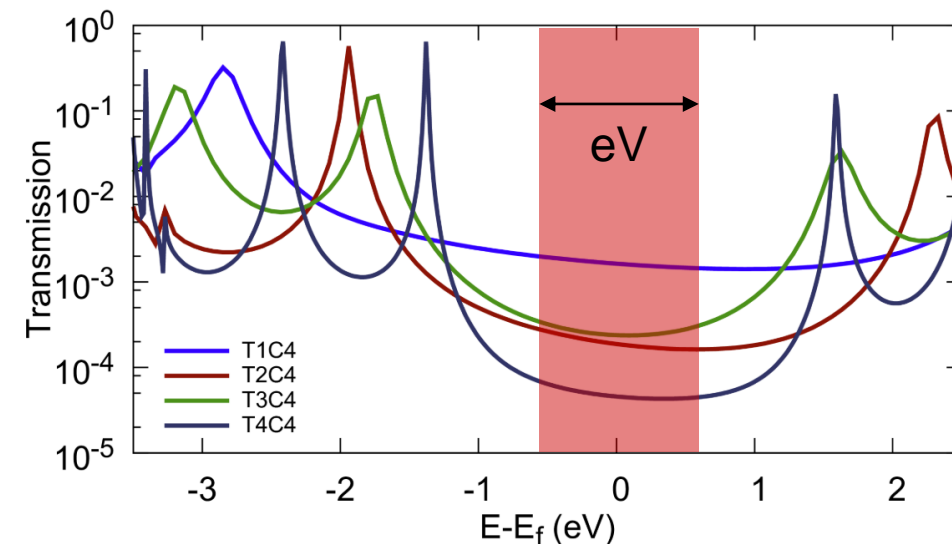
- > Single-molecular DFT calculations closer to CP AFM
- > Input: $T(E)$ from DFT on single-molecular junctions



$$I(V) = \frac{2e}{h} c \int_{E_F - eV/2}^{E_F + eV/2} T(E) \cdot dE$$

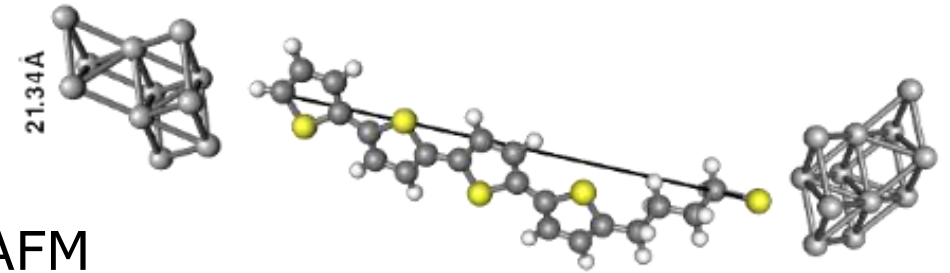
(Chem. Soc. Rev.2015,44, 875–888)

- > Integrating the area under the curve for an energy range of eV for every value of V



1) Simulating CP AFM I - V curves (DFT)

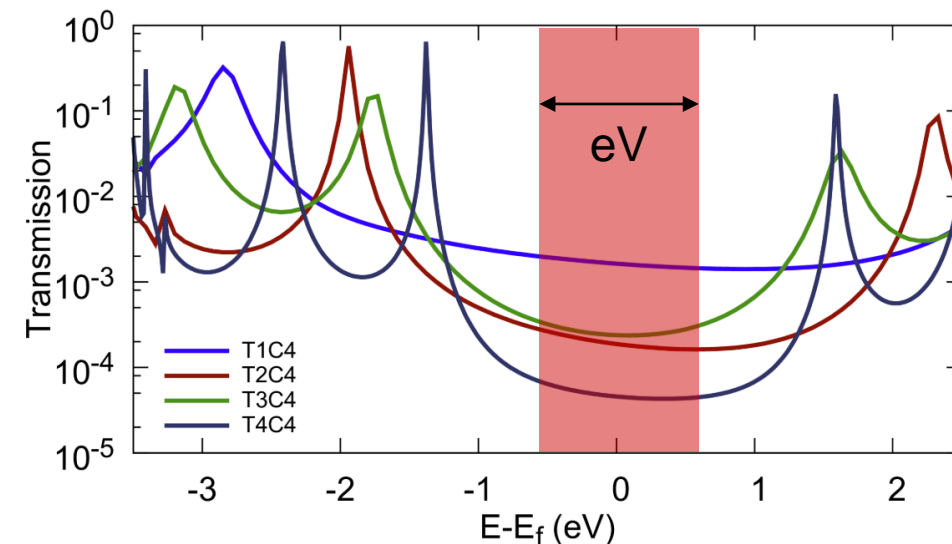
- Single-molecular DFT calculations closer to CP AFM
- Input: $T(E)$ from DFT on single-molecular junctions



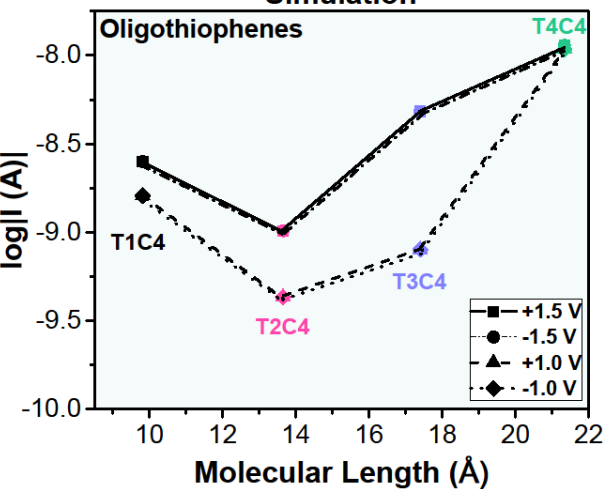
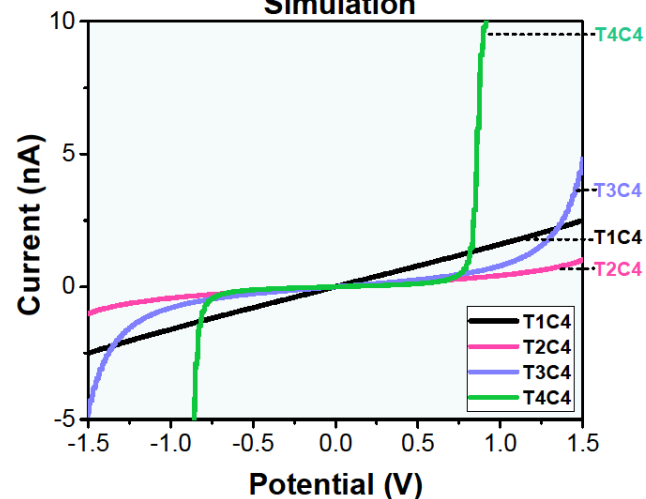
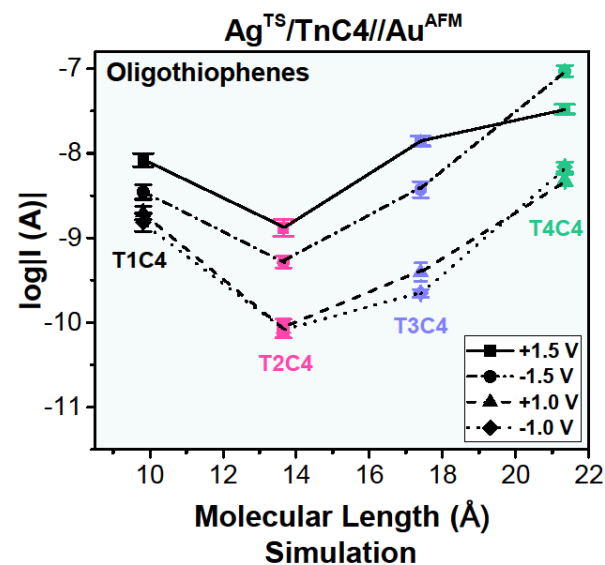
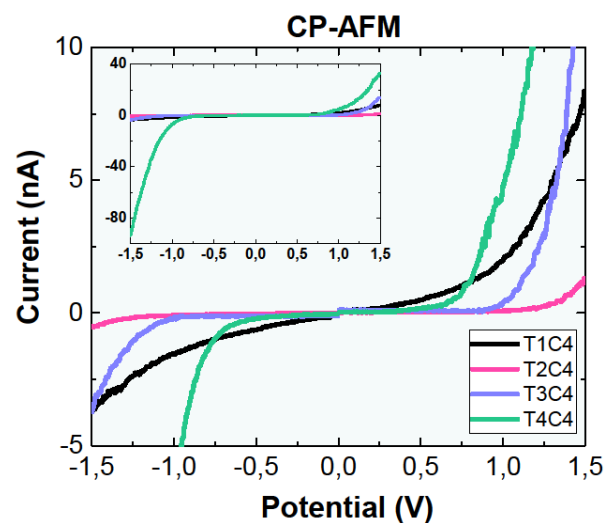
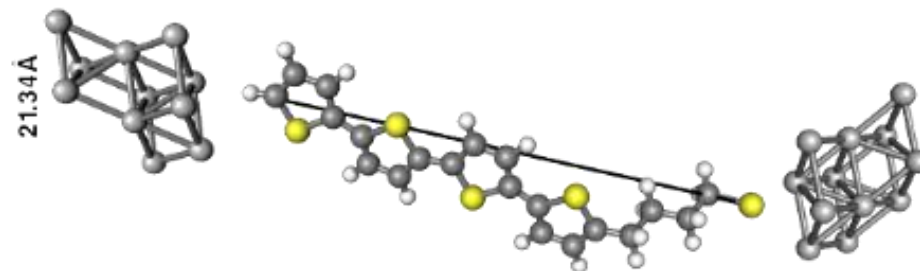
$$I(V) = \frac{2e}{h} c \int_{E_F - eV/2}^{E_F + eV/2} T(E) \cdot dE$$

(Chem. Soc. Rev.2015,44, 875–888)

- Integrating the area under the curve for an energy range of eV for every value of V
 - $E_f = -4.7\text{eV}$
 - $T(E)$ already contains molecules' DOS
 - Small constant (0.0001) added to $T(E)$

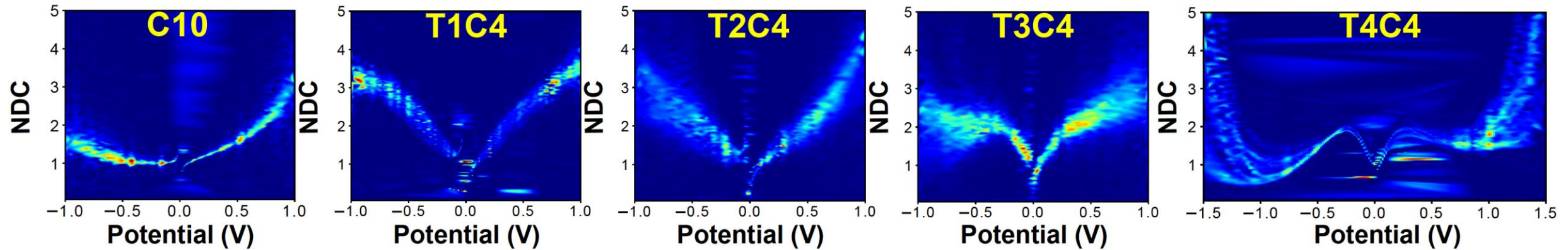


1) Simulating CP AFM I - V curves (DFT)



2) Simulating Normalised Differential Conductance (NDC)

Experiments



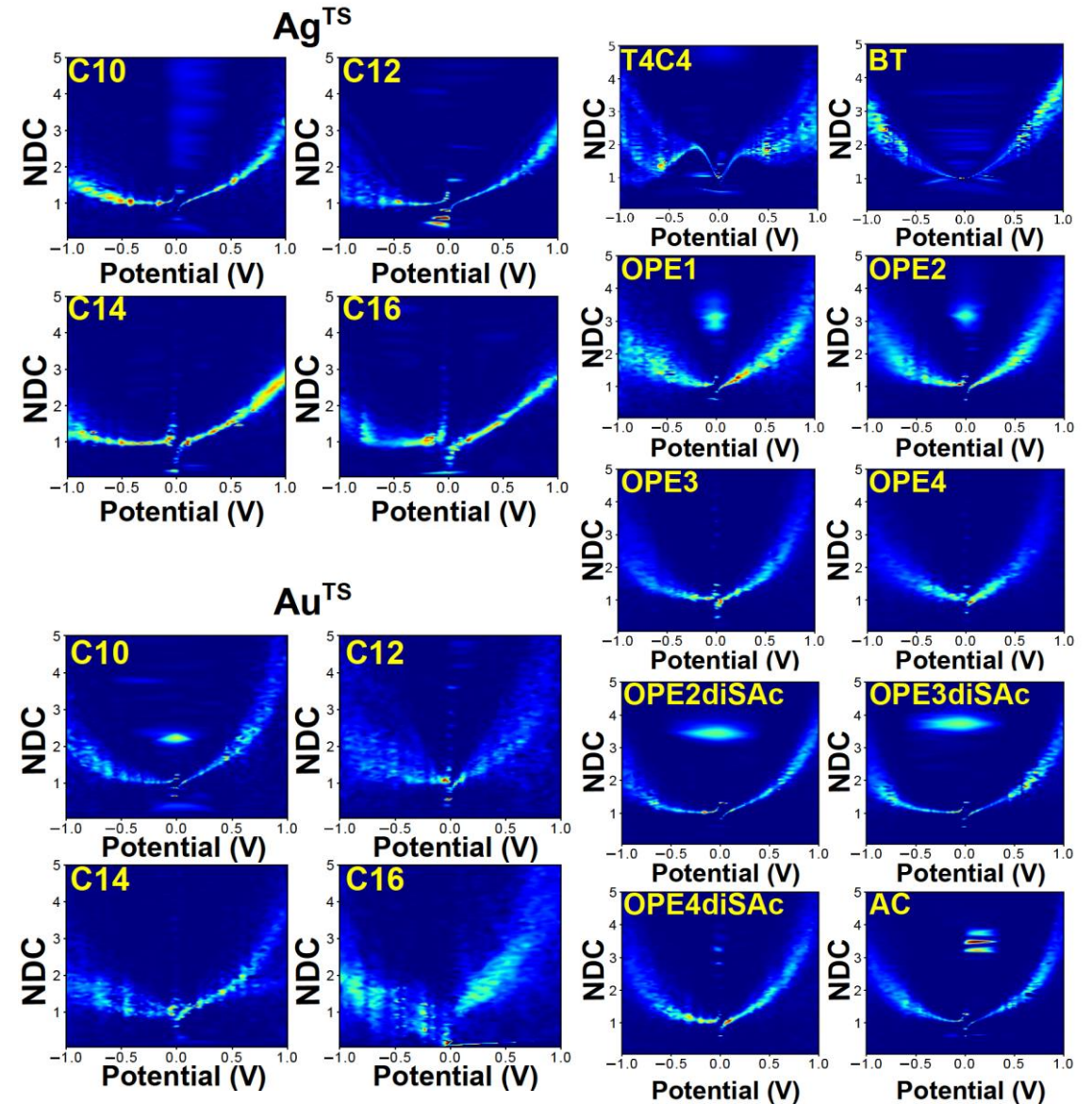
$$\text{NDC} = (dI/dV)/(I/V)$$

- > EGaIn measurements show bumps closing-in in the NDC curves as we go from T_1C_4 to T_4C_4

2) Simulating Normalised Differential Conductance (NDC)

$$\text{NDC} = (dI/dV)/(I/V)$$

- EGaIn NDC curves for aliphatic and conjugated molecules, otherwise do not show any bumps



2) Simulating Normalised Differential Conductance (NDC)

$$\text{NDC} = (dI/dV)/(I/V)$$

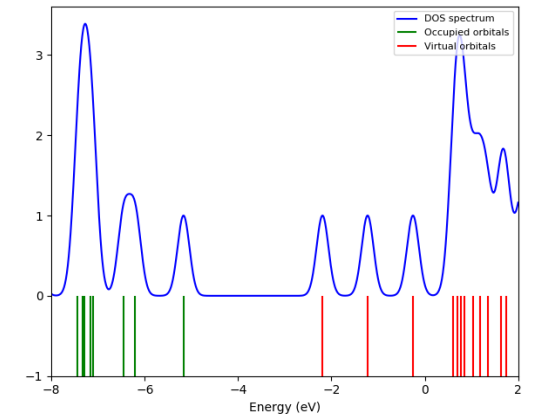
- › **Input:** DOS from DFT

DOS: Gaussian Fits to discrete energy levels of gas-phase single molecules

- › **Input:** $T(E)$ from STM equation

$$T(E, eV) = e^{-2d\sqrt{\frac{2m}{\hbar^2}\left(\phi - E + \frac{eV}{2}\right)}}$$

d is barrier width, ϕ is work function



2) Simulating Normalised Differential Conductance (NDC)

$$I(V) = c \int_{E_F - eV/2}^{E_F + eV/2} T(E, eV) \cdot DOS(E) \cdot dE$$

- › Integrating the area under the curve for an energy range of eV for every value of V

2) Simulating Normalised Differential Conductance (NDC)

$$I(V) = c \int_{E_F - eV/2}^{E_F + eV/2} T(E, eV) \cdot DOS(E) \cdot dE$$

- › Integrating the area under the curve for an energy range of eV for every value of V
 - $E_f = -4.7$ eV
 - Small constant (0.005) added to $DOS(E)$, so that it is not zero anywhere
 - Tunnelling barrier only comprises of the alkyl part

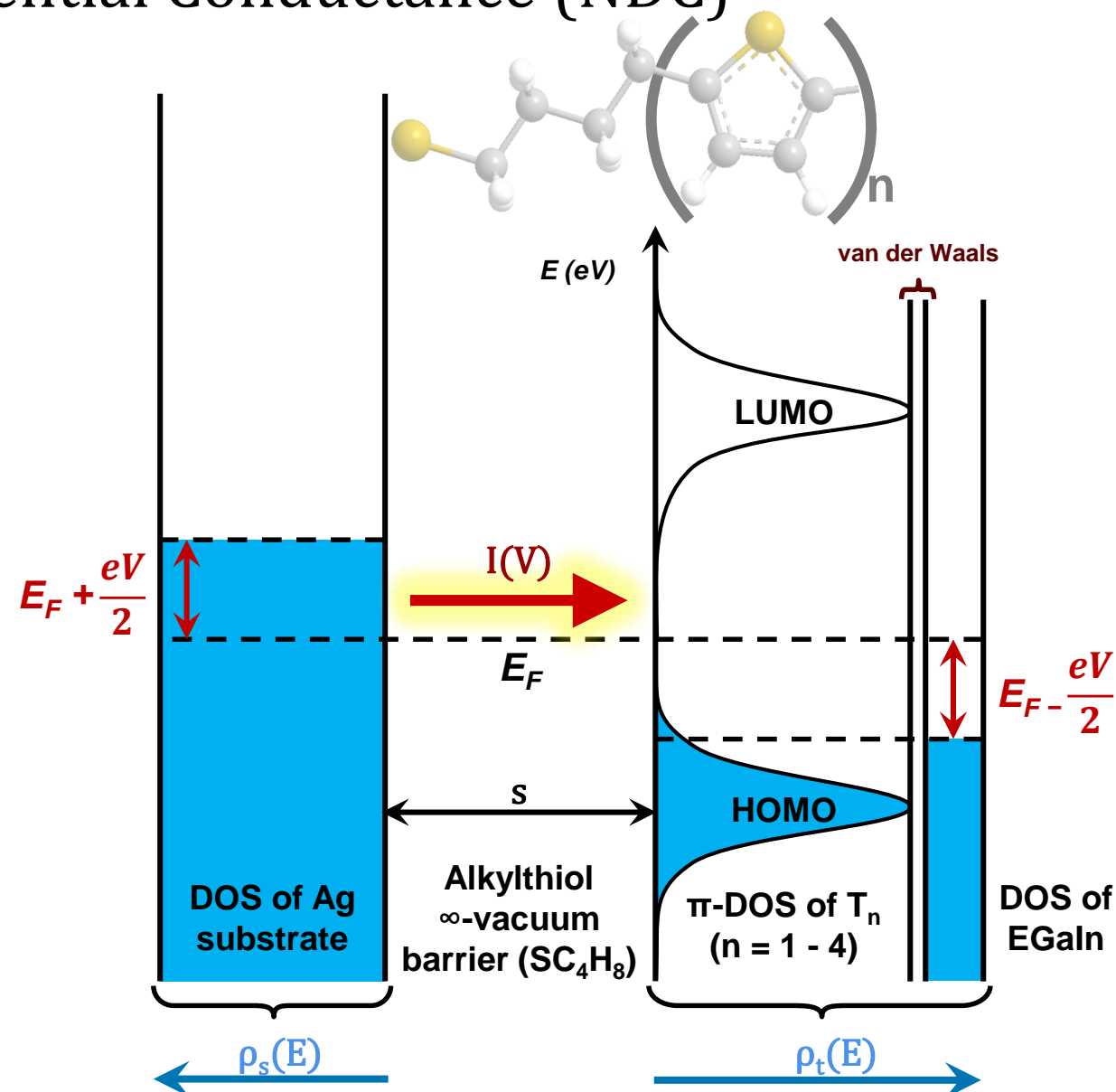
$$NDC = (dI/dV)/(I/V)$$

2) Simulating Normalised Differential Conductance (NDC)

Input 1: $DOS(E)$ from DFT Gaussian fits

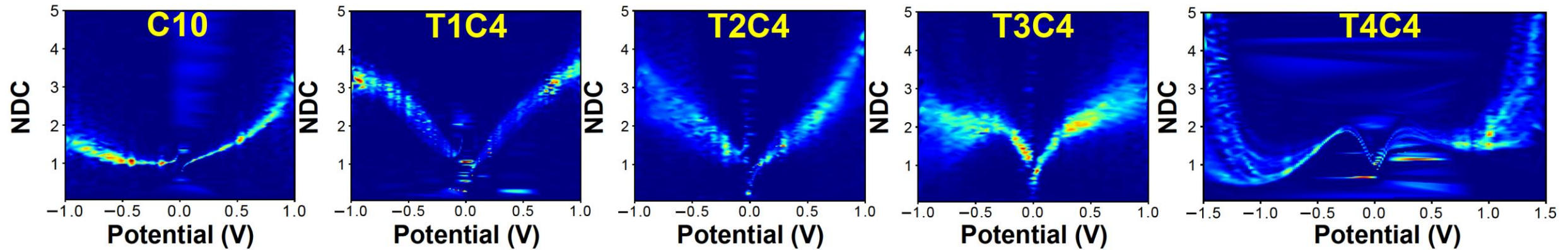
Input 2: $T(E, eV) = e^{-2d\sqrt{\frac{2m}{\hbar^2}\left(\phi - E + \frac{eV}{2}\right)}}$

$$I(V) = c \int_{E_F - eV/2}^{E_F + eV/2} T(E, eV) \cdot DOS(E) \cdot dE$$

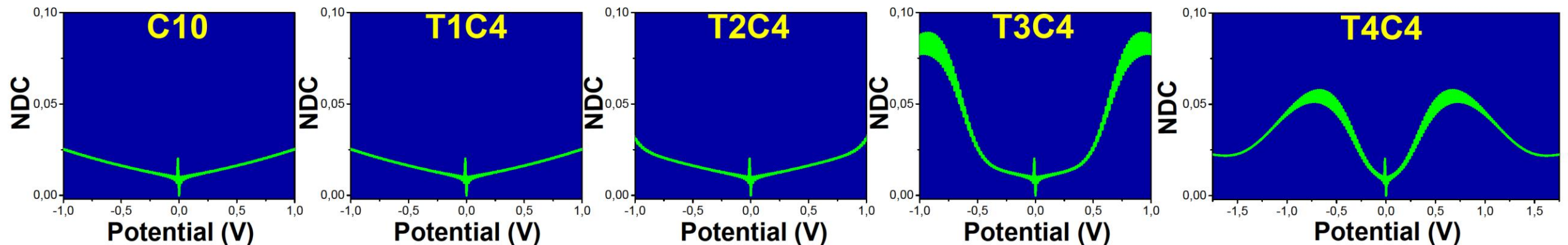


2) Simulating Normalised Differential Conductance (NDC)

A) Experiments



B) Simulations



Thank you for your attention!



✓ Cite This: *J. Am. Chem. Soc.* 2018, 140, 15048–15055

Article

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Tunneling Probability Increases with Distance in Junctions Comprising Self-Assembled Monolayers of Oligothiophenes

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